

MEL'NIKOV, N. N.

N. N. MEL'NIKOV, Ya. A. Mandel'baum and I. L. Vladimirova, Fertilizer and Insectofungicide Institute in Moscow

"Organic Insectofungicides. XII. Synthesis of Mixed Esters of Phosphoric and Thiophosphoric Acids", Zhur Obshchei Khim 23, 429-32 and 433-5 (1953)

All of the compounds listed in the paper are far below the insecticidal action of Parathion, which is just about the US standard at this current time. While it does not seem likely that these compounds are war agents, it actually cannot be said with certainty whether or not they are sufficiently toxic to humans to cause death or disability. The only Soviet chemical work dealing with compounds that have biological action analogous to the US and UK published material on the fluorophosphates (that is, anti-cholinesterase action) is the published research by Mel'nikov and co-workers on the synthesis and insecticidal properties of some esters of various phosphoric acids. As I have said before, the real significance of this type of work lies in the fact that it deals with anti-cholinesterases. Mel'nikov's published research in 1950 appearing in Doklady Akad Nauk SSSR 71, 485-7 is the first acknowledgment in the Soviet press of compounds related to Parathion, that is to say, compounds that are organophosphorous anti-cholinesterases.  
Digested translation available.

SO: B67003

Chemical Abst.  
Vol. 48 No. 8  
Apr. 25, 1954  
Organic Chemistry

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Use of the Willgerodt reaction in the synthesis of substituted phenyl- and naphthylacetic acids. V. A. Banks, *J. Org. Chem.*, 19, 132 (1954). The reaction of acetophenone with a substituted benzene ring in the presence of a catalyst and a base yields a substituted acetophenone which is then converted to a substituted thiophenol. The yield of acetophenone is affected by the nature of the substituent on the benzene ring. The following are the yields of acetophenone obtained from the reaction of acetophenone with a substituted benzene ring: p-Cl, 23 and 7; p-I, 12 and 0; p-Me, 63 and 0; o-HO, 0 and 0; p-HO, 0 and 0; p-MeO, 25 and 0; 2,5-Br, 0 and 0; 2,4-Cl, 39 and 5.5; 2,5-Cl, 34.7 and 0; 3-methyl-4-bromo, 53 and 0; 2,6-Br, 74.7 and 0; 2-methyl-4-chloro, 72.4 and 0; 3,6-isomer, 87.3 and 0; 3-methyl-2-chloro, 74.8; 2,4-(MeO), 0; 2,4,6-Cl, 0; 2,4,6-Me, 42.1; isomer, 61.3; 2,4-(MeO), 0; 2,4,6-Cl, 0; 2,4,6-Me, 42.1. Me 1-naphthyl ketones (substituents on naphthalene ring given): none 87.3 (acid, m. 132°; amide, m. 182°); 4-Br, 0; 4-Cl, 0; 4-Me, 89.3; 4-MeO, 20; 4-Et, 41; 4-EtO, 0; 4-Pr, 40; and 4-Bu, 34. O. M. Kosolapoff.

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Use of the Willgerodt reaction in the synthesis of substituted phenyl- and naphthylacetic acids. Yu. A. Baskakov and N. N. Mel'nikov. *J. Gen. Chem. U.S.S.R.* 23, 005-7(1953) (English translation). See *C.A.* 48, 4477d. H. L. H.

GALASHINA, M. L.; VLADIMIROVA, I. L.; MANDELBAUM, Ya. A.; YELNIKOV, N. N.

Insectifuges

Organic insectifuges. Part 13. Synthesis of mixed esters of phosphoric and thiophosphoric acids containing the simplest substituents in the aliphatic radical. Zhur. ob. khim. 23, No. 3, 1953.

Monthly List of Russian Accessions, Library of Congress, June 1953. Uncl.

USSR/Chemistry - Phosphorus Organic Compounds Aug 53

"Research in the Field of Organic Insecto-fungicides. XIV. Synthesis of Some Mixed Esters of Dithiophosphoric Acid," <sup>111</sup>N. N. Mel'nikov and K. D. Shvetsova-Shilovskaya, *Sci Inst of Fertilizers and Insecto-fungicides*

Zhur Obshch Khim, Vol 23, No 8, 1352-1357

Studied the reaction between dialkyldithiophosphoric acids and several unsatd compds. The compds react with the above acids to form mixed esters of dithiophosphoric acids with satisfactory yields.

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USSR/Chemistry - Phosphorus Organic Compounds Aug 53

"Research in the Field of Organic Insecto-fungicides.  
XV. Synthesis of Some Esters of Thiophosphoric Acid  
Containing Substituents in the Aromatic Radical,"  
N. N. Mel'nikov and D. N. Khokhlov

Zhur Obshch Khim, Vol 23, No 8, pp 1357-1364

Synthesized a series of esters of thiophosphoric acid in order to study their insecticidal properties. Established that ethyldichlorothiophosphate and diethylchlorothiophosphate can yield triarylthio-phosphates when treated with chloronitrophenolates.

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USER/Chemistry - Phosphorus Organic Compounds Sep 53

"Work in the Field of Organic Fungicides. XVI. Synthesis of Mixed Esters of Thiophosphoric Acid Containing Different Functional Groups in the Aromatic Radical," M.L. Galashina and N.N. Mel'nikov

Zhur Obshch Khim, Vol 23, No 9, pp 1539-1542

A number of dialkylalkoxy phenylthiophosphates, never before described in the literature, as well as diethyl-4-thiocyanophenylthiophosphate were

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synthesized for the purpose of studying insecticidal properties. This study showed that the above compounds are less active than diethyl-4-nitrophenylthiophosphate.

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The structure and the physiological activity of substituted phenylacetic and naphthylacetic acids. N. N. Melnikov, R. Kh. Purotskaya, Yu. A. Baskakov, A. N. Boyarkin, and M. S. Kuznetsova. *Doklady Akad. Nauk S.S.S.R.* 89, 853-8 (1953).—Introduction of halogen into the ring of  $\text{PhCH}_2\text{CO}_2\text{H}$  increases the physiol. activity, which is greatest with Cl; Br is less active; iodine in the  $\rho$ -position does not increase activity. Most active are the 2-halo, least active the 4-halo deriva., while the 3-halo deriva. are intermediate. This is the reverse order of activity in comparison with phenoxyacetic acids. Introduction of a 2nd halogen does not change the physiol. activity, while a 3rd halogen can either reduce or increase activity. A Me group in the  $\rho$ -position reduces the activity, and 3 Me groups reduce it very much. A smaller decrease in activity is caused by a  $\rho$ -MeO group. Halogenation of  $m$ -MeC<sub>6</sub>H<sub>4</sub>-OCH<sub>2</sub>CO<sub>2</sub>H causes increased activity, the greatest activity resulting from 4-substitution, and lesser activity from 2-halogenation. With 2,4-D as the standard which gave 100% increase of wheat coleoptile growth at 1 mg./l. concn., and heteroraxin as the standard which gave 100% increase of kidney bean root growth, the following physiol. results were obtained in plant tests; (in above order, with 1, 10, and 100 mg./l. concns. tested for the wheat growth, and 10, 50 and 100 mg./l. concns. for the kidney bean growth). Substituted phenylacetic acids (substituents, given): H, m.

|      |      |       |       |       |                              |                       |                      |   |
|------|------|-------|-------|-------|------------------------------|-----------------------|----------------------|---|
| 76°  | -2%  | +45%  | -     | +4%   | 0%                           | $\alpha$ -Cl, m. 65°  | +70%                 |   |
| +84% | -    | +6%   | +150% | -     | -                            | $m$ -Cl, m. 74°       | +9%                  |   |
| +91% | -    | -     | +62%  | +280% | -                            | $\rho$ -Cl, m. 106°   | +13%                 |   |
| +71% | -    | -     | +55%  | +91%  | -                            | $\alpha$ -Br, m. 104° | +7%                  |   |
| +71% | -    | -     | +33%  | +104% | -                            | $\rho$ -Br, m. 114°   | +2%                  |   |
| +80% | -    | -     | 0%    | +8%   | $\rho$ -I, m. 135°           | -4%                   | -8%                  |   |
| -    | -    | 0%    | +21%  | -     | 2,4-di-Cl, m. 131°           | +10%                  | +80%                 |   |
| -    | +68% | +162% | -     | -     | 2,5-di-Cl, m. 105°           | +13%                  | +57%                 |   |
| -    | -    | +30%  | +120% | -     | 2,3,4-tri-Cl, m. 136°        | +40%                  | -                    |   |
| +71% | -    | -     | +150% | -60%  | 2,4,5-tri-Cl analog, m. 128° | 0%                    | +58%                 |   |
| 128° | 0%   | +58%  | -     | +98%  | +100%                        | $\rho$ -MeO, m. 83°   | -                    |   |
| 83°  | -    | +20%  | +20%  | -     | 0%                           | $\rho$ -Me, m. 92°    | -                    |   |
| +5%  | +6%  | -     | 0%    | 0%    | 4,3-ClMe, m. 83°             | +22%                  | -                    |   |
| +73% | -    | -     | +7%   | +69%  | 4,3-BrMe, m. 81°             | -                     | -                    |   |
| +20% | +71% | -     | -     | 14%   | +60%                         | 6,3-ClMe, m. 86°      | -                    |   |
| -2%  | +40% | -     | -     | 0%    | -25%                         | 6,3-BrMe, m. 82°      | -                    |   |
| -8%  | +59% | -     | -     | +63%  | +60%                         | 2,4-ClMe, m. 100°     | -                    |   |
| +13% | +44% | -     | -     | +12%  | +33%                         | 3,4-Me(MeO), m. 127°  | -                    |   |
| 117° | -    | +1%   | +34%  | -     | 0%                           | 2,4,5-tri-Me, m. 107° | -                    |   |
| 107° | -    | +3%   | -8%   | -     | -21%                         | -25%                  | 2,3-Me(MeO), m. 107° | - |
| -    | -    | -12%  | +25%  | -     | 0%                           | 0%                    | -                    |   |

In substituted naphthylalkylcarboxylic acids it was shown that removal of CO<sub>2</sub>H from the ring by more than 1 C atom reduces the physiol. activity and the compds. with an acid group at the 1-position on C<sub>10</sub>H<sub>7</sub> skeleton are the only active ones; substances with 2-substitution are inactive. Substitution of alkyl groups or MeO groups into the C<sub>10</sub>H<sub>7</sub> ring leads to sharp decline in activity. The following results were obtained in biol. tests (same standards as above): Substituted 1-naphthaleneacetic acids (substituents given): H, m. 131°, +77%, +169%, - , +92%, - , 4-MeO, m. 144°, +2%, -12%, - , +10%, +27%, +9%, 4-Me, m. 148°, +1%, +1%, +4%, +36%, +24%, -10%; 4-Et, m. 120°, +9%, - , +9%, +6%, +27%; 4-Pr, m. 119°, -5%, -7%, - , +24%, +50%, +33%; 4-Bu, m. 110°, -14%, - , +21%, +20%, +33%. 1-C<sub>6</sub>H<sub>4</sub>COCH<sub>2</sub>CH<sub>2</sub>CO<sub>2</sub>H, m. 132°, +9%, +60%, - , - , - , 1-C<sub>10</sub>H<sub>7</sub>(CH<sub>2</sub>)<sub>2</sub>CO<sub>2</sub>H, m. 160°, +9%, +40%, - , - , - , 2-C<sub>10</sub>H<sub>7</sub>COCH<sub>2</sub>CH<sub>2</sub>CO<sub>2</sub>H, m. 174°, -25%, -9%, - , - , - , 2-C<sub>10</sub>H<sub>7</sub>(CH<sub>2</sub>)<sub>2</sub>CO<sub>2</sub>H, m. 109°, -3%, -5%, - , - , - . Most of the above acids were prepd. by the Willgerodt reaction others through the nitriles from the corresponding benzyl halides. G. M. Kosolapoff

Structure and physiological activity of some substituted phenoxyalkylcarboxylic acids on plants. N. N. Mel'nikov, R. Kh. Turetskaya, and K. S. Bokarev (K. A. Timiryazev Inst. Plant Physiol., Moscow). *Doklady Akad. Nauk S.S.S.R.* 90, 921-3(1963). — Tests with a large no. of substituted phenoxyalkylcarboxylic acids showed that an RO group attached to the Ph nucleus slightly raised the growth rate (kidney-bean cutting) and that a MeO gave a variable effect, the meta position giving the highest activity and the 2-position giving the lowest; the same is true for halogen substitution. An Et group in the 2-position of the acid radical raises the activity drastically, allyloxy derivs. are less active than the propoxy derivs., except for 2-allyloxy-phenoxyacetic acid, which is quite active. Amides may be more or less active than the free acids; anilides of allyloxy-phenoxyacetic acids are more active than the amides. 4-Methoxy-2-chlorophenoxyacetic acid is nearly as active as 3-indolebutyric acid, 4-methoxy-2-bromophenoxyacetamide is also highly active. G. M. Kosolapoff

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*MEL'NIKOV, N. N.*

Mel'nikov, N. N., Baskakov, Yu. A., and Bokarov, K. S.:  
Khimiya gerbitsidov i stimulyatorov rosta rastenii (Chem-  
istry of Herbicides and Stimulators of Plant Growth).  
Moscow: Goskhimizdat, 1956. 28 pp. Reviewed in *Fiziol.*  
*Rastenii* 2, No. 6, 689-90 (1955).

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MEL'NIKOV, N.N.; NABOKOV, V.A.; POKROVSKIY, Ye.A.; ZITSER, A.I., redaktor;  
YEVDOKIMOVA, Z.N., tekhnicheskij redaktor.

[DDT; properties and use] DDT; svoistva i primeneniye. Moskva, Gos. nauch-  
no-tekhn. izd-vo khimicheskoi lit-ry, 1954. 203 p. (MLRA 8:1)  
(DDT (Insecticide))

MEL'NIKOV, N. N.

Chemistry of herbicides and plant growth stimulators. Moskva, Gos. nauchnotekhn. izd-vo khim. promyshl., 1954. 381 p.

1. Herbicides. 2. Growth promoting substances.

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NAMETKIN, Sergei Semenovich, akademik; TOPCHIYEV, A.V., akademik, otvetstvennyy redaktor; MEL'NIKOV, N.N., doktor khimicheskikh nauk, zamestitel' otvetstvennogo redaktora; BRUSOV, I.I., redaktor; POLYAKOVA, T.V., tekhnicheskii redaktor

[Collected works] Sobranie trudov, Moskva, Izd-vo Akademii nauk  
SSSR. Vol. 1. 1954. 823 p. (MLRA 8:1)  
(Chemistry, Organic)

MeL'NIKOV, N. N.

Analysis of technical grade of 1-naphthaleneacetic acid and its methyl ester. Yu. A. Bakukay and N. N. Mel'nikov. *Khim. Prom.* (1954), 111-113. The methyl ester of the acid and its salts are used to spray fruit trees to retard premature blossoming and the loss of fruit and to spray cotton plants to increase the cotton yield, etc., while the methyl ester, 3.5% dusted on ground clay, is used to prevent the sprouting of potatoes during long storage. Pure 1-naphthaleneacetic acid is a white, cryst. substance, m. 132°, sol. in H<sub>2</sub>O; 0.4 g./l. at 20°, and readily sol. in org. solvents. The tech. grade may contain the 1,5-naphthaleneacetic acid, inorg. salts, naphthalene, and water. Me 1-naphthaleneacetate is an almost odorless, colorless liquid, b.p. 182-4°, d<sub>4</sub><sup>20</sup> 1.1459, n<sub>D</sub><sup>20</sup> 1.5975. The monoacid and its ester are used in ants, practically harmless to warm-blooded animals. The analysis is based on the extrn. of a water soln. of the acid with benzene and its titration with 0.05N NaOH. The residual soln. from the benzene extrn. is then extrd. with ether, and the 1,5-naphthaleneacetic acid is titrated with 0.05N NaOH.

W. M. Sternberg

Мел'ников, Н. Н.

✓ Analysis of herbicidal preparations. K. S. Bokarev and N. N. Mel'nikov. *Khim. Prom.* 1954, No. 1, 42-3.—*Detn.* of 2,4-dichlorophenoxyacetic acid and of its BuOH ester, carried out by titration with NaOH with methyl red as indicator avoids the error caused by the usually present 2,4-dichlorophenol and gives good results. Procedures for detn. of the 2 weed killers were worked out. 2,4-Dichlorophenoxyacetic acid is extd. from a dissolved and acidified sample with Et<sub>2</sub>O, the ether is evapd., and the residue is dissolved in neutral alc., and titrated with 0.1N NaOH with methyl red as indicator. To det. free 2,4-dichlorophenoxyacetic acid in its Bu ester, a sample of the ester is dissolved in neutral EtOH, dild. with H<sub>2</sub>O, and titrated with NaOH with methyl red as indicator. To det. the ester, a sample is refluxed with alc. KOH. The alc. is evapd., the residue is dissolved in acid. KCl soln., the soln. is extd. with Et<sub>2</sub>O, the ether is evapd., and the residue is dissolved in EtOH and titrated with NaOH with methyl red as indicator.

M. Hosh.

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USSR

2527. Analysis of technical 1-naphthylacetic acid and its methyl ester. Yu. A. ~~Melnikov~~ and N. N. Melnikov (Khim. Prom-st., 1964, 12, 4754; ~~Reperativnyi Zh., Khim., 1964, Abstr. No. 43,621~~).  
 A sample of technical methyl 1-naphthylacetate is saponified with KOH soln.; the acid is extracted from the acidified saponification mixture with benzene and determined by adding water to the benzene soln. and titrating with 0.05 N KOH soln. at 70° to 75° C. 1:5-Naphthalenediacetic acid is then extracted from the acidified saponification mixture with ether; the ether is removed by distillation, the residue is dissolved in 80 per cent ethanol, and the acid is determined by titration with 0.05 N KOH soln. E. Havgs

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MEL'NIKOV, N. N.

Note on the paper by Ivanova "Quantitative determination of Parathion." N. N. Mel'nikov. *Gigiena i Sanit.* 1954, No. 6, 63. — The method suggested by I. (C.A. 47, 11046g) is unworkable since this method detcs. not (EtO)P(OC<sub>6</sub>H<sub>4</sub>NO<sub>2</sub>-p) (I) but the impurity represented by p-nitrophenol, which in a com. product is usually not over 10-20%. The rate of hydrolysis of I by NaOH at 25° has the const. 0.047, so that hydrolysis at 20° would require several days and not the 20 min. stated by I. It is suggested that the mixt. be refluxed 1 hr., but the method still does not give good results. The best method is based on reduction of I, diazotization of the amine, and coupling this with some suitable coupling agent. G. M. K.

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USSR/Chemistry - Agricultural

FD-868

Card 1/1 Pub.50 - 1/24

Author : Vol'fkovich, S. I., Mel'nikov, N. N., Orlov, V. I.

Title : The chemical industry in the fight to increase yields and preserve crops (Concerning the opening of the All-Union Agricultural Exposition).

Periodical : Khim. prom., No. 6, 321-331 (1-11), Sep 1954

Abstract : Review general trends in USSR agricultural chemistry and current production plans and other developments in fertilizers, insecticides, fungicides, herbicides, and plant growth stimulants. Six references, all USSR, all since 1940. Three figures.

Institution :

Submitted :

USSR MOSCOW, U.S.S.R.

AID P - 264

Subject : USSR/Chemistry  
Card : 1/1  
Authors : Mel'nikov, N. N. and Baskakov, Yu. A. (Moscow)  
Title : Chemical agents for weed control and plant growth substances  
Periodical : Usp. khim. 23, No. 2, 142-198, 1954  
Abstract : Review of various herbicides and growth promoting substances. The physiological activity of some hydrocarbons, alcohols, phenols, ethers, aldehydes and ketones, organic acids, amines, and heterocyclic compounds is discussed. 685 references (175 Russian): 1850-1953.  
Institution : None  
Submitted : No date

MEL'NIKOV, N. N.

USSR/Chemistry - Biochemistry

Card 1/1 Pub. 151 - 36/38

Authors : Baskakov, Yu. A., and Mel'nikov, N. N.

Title : Synthesis and physiological activity in plants of isopropyl esters of certain arylcarbamic acids

Periodical : Zhur. ob. khim. 24/2, 376-379, Feb 1954

Abstract : Numerous hitherto unknown isopropyl esters of arylcarbamic acid were synthesized and their physiological activities in young wheat plants were investigated. It was established that the substitution of the hydrogen in the phenyl radical of phenylisopropylcarbamate considerably affects the activity of the compound. Experiments showed that para- and ortho-isomers are much less active than corresponding meta-isomers. The introduction of functional groups into the phenyl radical and its effect on the physiological activity of these compounds in plants are explained. Fifteen references: 6-USA; 8-USSR and 1-English (1916-1953). Tables.

Institution : Academy of Sciences USSR, The K. A. Timiryazev Institute of Plant Physiology

Submitted : September 15, 1953

MEL'N'KOV, N. N.

U.S.S.R.

*Synthesis and physiological activity on plants of isopropyl  
esters of some erythroester acids* — Ya. A. Buzukov and  
N. N. Mel'nikov. *Izv. Gos. Chern. U.S.S.R.* 24, 385-9  
(1954) (Engl. translation). — See C.A. 48, 8460a.

H. L. H. J.

MEL'NIKOV, N.N.

USSR/ Chemistry      Synthesis methods

Card                : 1/1      Pub. 151 - 25/35

Authors            : Baskakov, Yu. A., and Mel'nikov, N. N.

Title                : Synthesis of cyclic hydrazide of maleic acid and some of its derivatives

Periodical         : Zhur. ob. khim. 24, Ed. 7, 1216 - 1221, July 1954

Abstract           : The method of synthesizing cyclic hydrazide of maleic acid and several of its derivatives, is described. Data are presented on hitherto unknown substituted cyclic hydrazides of maleic acid obtained from the maleic anhydride derivatives. The mechanism of reaction, leading to the formation of 1,2-dihydropyridazinedione-3,6 and its analogues from anhydrides and hydrazine salts, is explained. Nine USA, 2 German and 2 USSR references.

Institution        : Acad. of Sc. USSR, The K. A. Timiryazev Institute of Plant Physiology

Submitted         : February 4, 1954

Mel'nikov, N.N.

Syntheses and physiological activity in plants of some 2-aryloxyethanols and aryloxyacetones. K. S. Bokarev and N. N. Mel'nikov (K. A. Timiryazev Inst. Plant Physiol. Acad. Sci. U.S.S.R., Moscow). *Zhur. Obshch. Khim.* 24, 2014-23 (1954).—Various aryloxyethanols were prepd. by refluxing ArOH and ClCH<sub>2</sub>CH<sub>2</sub>OH in aq. NaOH 2 hrs. followed by extr. with (CH<sub>2</sub>Cl)<sub>2</sub>. Thus were prepd. the following ArOCH<sub>2</sub>CH<sub>2</sub>OH (An. % yield, and phys. consts. given): 4-BrC<sub>6</sub>H<sub>4</sub>, 63.9, m. 65°; 2,4,5-Cl<sub>3</sub>C<sub>6</sub>H<sub>3</sub>, 80, m. 65°; 63.2, m. 58°, b<sub>10</sub> 185-90°; 2,4,5-Cl<sub>3</sub>C<sub>6</sub>H<sub>3</sub>, 80, m. 65°; 2-MeOC<sub>6</sub>H<sub>4</sub>, 63.2, b<sub>10</sub> 163-4°, d<sub>20</sub> 1.1561, n<sub>D</sub><sup>20</sup> 1.5430; 3-isomer, 79, b<sub>10</sub> 143°, d<sub>20</sub> 1.1531, n<sub>D</sub><sup>20</sup> 1.5416; 2-EtOC<sub>6</sub>H<sub>4</sub>, 65, m. 38-9°, b<sub>10</sub> 129-30°; 3-PrOC<sub>6</sub>H<sub>4</sub>, 83.6, b<sub>10</sub> 157°, d<sub>20</sub> 1.0867, n<sub>D</sub><sup>20</sup> 1.5240; 4-isomer, 70.5, m. 75°; 3-BuOC<sub>6</sub>H<sub>4</sub>, 84.3, b<sub>10</sub> 170-2°, d<sub>20</sub> 1.0501, n<sub>D</sub><sup>20</sup> 1.5197; 2,4-Cl<sub>2</sub>(PrO)C<sub>6</sub>H<sub>3</sub>, 61.7, b<sub>10</sub> 169-71°, d<sub>20</sub> 1.1920, n<sub>D</sub><sup>20</sup> 1.5367; 2,4-Cl<sub>2</sub>(BuO)C<sub>6</sub>H<sub>3</sub>, 64.5, b<sub>10</sub> 171-2°, d<sub>20</sub> 1.1631, n<sub>D</sub><sup>20</sup> 1.5309; 2,4-Cl<sub>2</sub>(iso-AmO)C<sub>6</sub>H<sub>3</sub>, 64.8, b<sub>10</sub> 191-4°, d<sub>20</sub> 1.1310, n<sub>D</sub><sup>20</sup> 1.5241; 2,4-Br(BuO)C<sub>6</sub>H<sub>3</sub>, 66.7, b<sub>10</sub> 200-2°, d<sub>20</sub> 1.3242, n<sub>D</sub><sup>20</sup> 1.5433; 4,5,2-Cl<sub>3</sub>(MeO)C<sub>6</sub>H<sub>3</sub>, 66.7, m. 94°; 2-EtO analog, 44.8, m. 77°; 2-PrO analog, 73, m. 55°, b<sub>10</sub> 177-9°; 2-BuO analog, 71.2, b<sub>10</sub> 190-2°, d<sub>20</sub> 1.2629, n<sub>D</sub><sup>20</sup> 1.5460; 2-iso-AmO analog, 62.8, b<sub>10</sub> 160-52°, d<sub>20</sub> 1.1918, n<sub>D</sub><sup>20</sup> 1.5350; 4,5,2-Br(BuO)C<sub>6</sub>H<sub>3</sub>, 71.8, m. 112°; 2-BuO analog, 57, m. 88°; 2-PrO

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sons, m. 181-2°; 2,4-Cl<sub>2</sub>C<sub>6</sub>H<sub>3</sub>, 51, m. 54°, b. 138-41°  
 (semicarbazone, m. 184-5°); 2,4,5-Cl<sub>3</sub>C<sub>6</sub>H<sub>2</sub>, 57.3, m. 116°  
 (semicarbazone, m. 202°); Cl<sub>2</sub>C<sub>6</sub>, 80, m. 105-5.5° (semi-  
 carbazone, m. 193-4°, decampn.); 2-BrC<sub>6</sub>H<sub>4</sub>, 50.2, b. 150-8°  
 (semicarbazone, m. 150-1°); 2,4-Cl(FrO)C<sub>6</sub>H<sub>3</sub>,  
 51, m. 60°, b. 171-2° (semicarbazone, m. 147°); 4-BuO  
 analog, 49, b. 178-5° (semicarbazone, m. 163°); 4,5,2-  
 Cl(EtO)C<sub>6</sub>H<sub>3</sub>, 73.6, m. 64° (semicarbazone, m. 168-9°);  
 2-BuO analog, 52.0, m. 69°, b. 155-8° (semicarbazone,  
 m. 172°); 4,5,3-Br(MeO)C<sub>6</sub>H<sub>3</sub>, 68.3, m. 110° (semicar-  
 bazone, m. 173-4°); 4,6,3-Br(EtO)C<sub>6</sub>H<sub>3</sub>, 65.8, m. 72°  
 (semicarbazone, m. 178°); 2-PrO analog, 84.5, m. 71°  
 (semicarbazone, m. 188.5°); 2-BuO analog, 81.7, m. 81°  
 (semicarbazone, m. 172-3°); 2-iso-AmO analog, 73.6,  
 b. 198-201° (semicarbazone, m. 171.5°); 4-C<sub>6</sub>H<sub>4</sub>SCH<sub>2</sub>Ac,  
 79.6%, m. 135° (semicarbazone, m. 198°) (cf. Stoermer, Ann.  
 312, 237(1900)]. 4-BrC<sub>6</sub>H<sub>4</sub>OCH<sub>2</sub>Ac semicarbazone, m. 197°;  
 2,4,6-tri-Br analog, m. 198.5°. (2-Benzothiazolylthio)acetone  
 semicarbazone m. 152-3°.  
 G. M. Kosolapoff

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*N. S. BAKAREV*

$C_8H_8N_2$ , m. 181-2°; 2,4- $C_8H_8N_2$ , 54, m. 54°, b. 138-41°  
 (semicarbazone, m. 184-5°); 2,4,5- $C_8H_8N_2$ , 57.3, m. 116°  
 (semicarbazone, m. 202°);  $C_8H_8N_2$ , 80, m. 105-5.5° (semi-  
 carbazone, m. 193-4°, decomp.); 2-Br- $C_8H_8N_2$ , 50.2, b. 150-5° (semicarbazone, m. 150-1°); 2,4- $C_8H_8N_2$ , 51, m. 60°, b. 174-5° (semicarbazone, m. 147°); 4-BuO analog, 49, b. 174-5° (semicarbazone, m. 163°); 4,5,2- $C_8H_8N_2$ , 73.6, m. 64° (semicarbazone, m. 168-9°); 2-BuO analog, 69.0, m. 69°, b. 155-8° (semicarbazone, m. 172°); 4,5,3-Br( $MeO$ ) $C_8H_8N_2$ , 68.3, m. 110° (semicarbazone, m. 173-4°); 4,5,3-Br( $EtO$ ) $C_8H_8N_2$ , 68.6, m. 72° (semicarbazone, m. 178°); 2-PrO analog, 94.6, m. 71° (semicarbazone, m. 166.5°); 2-BuO analog, 81.7, m. 61° (semicarbazone, m. 172-3°); 2-iso-AmO analog, 73.6, b. 198-200° (semicarbazone, m. 171.5°); 2- $C_8H_8N_2SCH_2Ac$ , 79.6%, m. 135° (semicarbazone, m. 196°) (cf. Stoermer, Ann. 312, 237(1900)); 4-Br $C_8H_8N_2OCH_2Ac$  semicarbazone m. 197°; 2,4,6-tri-Br analog, m. 196.5°. (2-Benzothiazolylthio)acetone semicarbazone m. 182-3°.

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G. M. Kosolapoff

МЕЛНИКОВ, Н.Н.

Application of products of organic synthetic industry in agriculture: N. N. Melnikov. *Zhur. Priklad. Khim.* 27, 577-93(1954).—A review with 261 references covering the agricultural uses of hydrocarbons, halogen derivs., nitro compds., alcs., phenols, carbonyl compds., org. acids and their derivs., compds. of S, P, Hg, and As, and pyrethrins. G. M. Kosolapoff —

MELNIKOV, N.N.

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Structures and insecticidal activity of some mixed esters of dithiophosphoric acid. K. A. Gar, N. N. Mel'nikov, Yu. N. Fadeev, and K. D. Shvetsova-Shimovskaya. *Zhurnal Akad. Nauk S.S.S.R.* 94, 241-4 (1954); cf. *Zhur. Obshchei Khim.* 23, 1352 (1953).—Insecticidal tests against *Calandra oryzae* with the following dithiophosphates indicated that mixed aliphatic aromatic esters are relatively weakly active and the activity drops with increase of the aliphatic radicals of the ester; hydroxyalkyl, acetoxyalkyl, and aldehydoalkyl esters have very low activity. The activity of trialkyl dithiophosphates rises significantly when a H atom of the alkyl group is replaced by a group like CN or CO<sub>2</sub>R. The following esters are reported (b., d., %<sup>12</sup>, concn. in % giving 50% mortality after 3 days exposure to aq. emulsion): (MeO)<sub>2</sub>P<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>Ph, b<sub>2</sub> 128-32°, 1.2102, 1.5705, 0.1; di-Et ester, b<sub>2</sub> 135-7°, 1.1444, 1.5498, 0.118; di-Pr ester, b<sub>2</sub> 120-31°, 1.01990, 1.5381, >1; di-iso-Pr ester, b<sub>2</sub> 121-4°, 1.0986, 1.5305, >1; di-Bu ester, b<sub>2</sub> 127-40°, 1.0890, 1.5320, nontoxic; di-iso-Bu ester, b<sub>2</sub> 117-22°, 1.0860, 1.5301, nontoxic; (BuO)<sub>2</sub>P<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>OH, b<sub>2</sub> 90-5°, 1.0491, 1.4835, over 0.5; (MeO)<sub>2</sub>P<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>OAc, b<sub>2</sub> 100-2°, 1.1558, 1.5255, 4.7; di-Et ester, b<sub>2</sub> 115-17°, 1.1517, 1.4948, 1.68; di-Pr ester, b<sub>2</sub> 80-1°, 1.0984, 1.5076, nontoxic; di-iso-Pr ester, b<sub>2</sub> 72-4°, 1.0846, 1.4336, nontoxic; di-Bu ester, b<sub>2</sub> 109°, 1.0915, 1.4858, nontoxic; di-iso-Bu ester, b<sub>2</sub> 104°, 1.0986, 1.4916, nontoxic; (iso-PrO)<sub>2</sub>P<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>CHO, b<sub>2</sub> 74°, 1.1248, 1.5095, nontoxic; di-Bu ester, b<sub>2</sub> 75-7°, 1.0756, 1.4955, nontoxic; (EtO)<sub>2</sub>P<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>CN, b<sub>2</sub> 134-42°, 1.1704, 1.5195, 0.03; di-Pr ester, b<sub>2</sub> 110-20°, 1.0506, 1.5068, 0.25; di-iso-Pr ester, b<sub>2</sub> 93-9°, 1.0182, 1.5020, over 0.25; di-Bu ester, b<sub>2</sub> 121-3°, 1.0816, 1.5050, 0.36; di-iso-Bu ester, b<sub>2</sub> 122-3°, 1.0986, 1.5010, over 0.30; (MeO)<sub>2</sub>P<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>CO<sub>2</sub>Me, b<sub>2</sub> 145°, 1.2025, 1.5160, over 0.3; di-Et analog, b<sub>2</sub> 167°, 1.1911, 1.5050, 0.084; di-iso-Pr analog,

b<sub>2</sub> 92-7°, 1.1420, 1.4918, over 0.5; di-iso-Bu analog, b<sub>2</sub> 124-9°, 1.1162, 1.4915, 0.6; (MeO)<sub>2</sub>P<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>MeCO<sub>2</sub>Me, b<sub>2</sub> 133-5°, 1.2330, 1.5100, 0.15; di-Et analog, b<sub>2</sub> 154.5°, 1.1577, 1.4995, over 0.15; di-iso-Pr analog, b<sub>2</sub> 84-5°, 1.1293, 1.4935, nontoxic; di-Bu ester, b<sub>2</sub> 110°, 1.1132, 1.4918, nontoxic; di-iso-Bu analog, b<sub>2</sub> 115-18°, 1.1138, 1.4915, nontoxic; (EtO)<sub>2</sub>P<sub>2</sub>CH<sub>2</sub>(CH<sub>2</sub>Ph)CO<sub>2</sub>Et, b<sub>2</sub> 134°, 1.1300, 1.5341, nontoxic; (MeO)<sub>2</sub>P<sub>2</sub>CH<sub>2</sub>(CO<sub>2</sub>Me)CH<sub>2</sub>CO<sub>2</sub>Me, b<sub>2</sub> 134.5°, 1.2804, 1.5070, 0.007-0.013; (MeO)<sub>2</sub>P<sub>2</sub>CH<sub>2</sub>(CO<sub>2</sub>Et)CH<sub>2</sub>CO<sub>2</sub>Et, b<sub>2</sub> 160-70°, 1.2076, 1.4960, 0.0033; (MeO)<sub>2</sub>P<sub>2</sub>CH<sub>2</sub>(CO<sub>2</sub>CHMe<sub>2</sub>)CH<sub>2</sub>CO<sub>2</sub>CHMe<sub>2</sub>, b<sub>2</sub> 123°, 1.1824, 1.4810, 0.003; (MeO)<sub>2</sub>P<sub>2</sub>CH<sub>2</sub>(CO<sub>2</sub>CH<sub>2</sub>CHMe<sub>2</sub>)CH<sub>2</sub>CO<sub>2</sub>CH<sub>2</sub>CHMe<sub>2</sub>, b<sub>2</sub> 127-8°, 1.1483, 1.4835, nontoxic; (EtO)<sub>2</sub>P<sub>2</sub>CH<sub>2</sub>(CO<sub>2</sub>Me)CH<sub>2</sub>CO<sub>2</sub>Me, b<sub>2</sub> 110-20°, 1.2237, 1.4979, 0.0016-0.0021; (EtO)<sub>2</sub>P<sub>2</sub>CH<sub>2</sub>(CO<sub>2</sub>Et)CH<sub>2</sub>CO<sub>2</sub>Et, b<sub>2</sub> 157-63°, 1.1742, 1.4910, 0.021; (EtO)<sub>2</sub>P<sub>2</sub>CH<sub>2</sub>(CO<sub>2</sub>CHMe<sub>2</sub>)CH<sub>2</sub>CO<sub>2</sub>CHMe<sub>2</sub>, b<sub>2</sub> 117-21°, 1.1493, 1.4815, 0.0034; (EtO)<sub>2</sub>P<sub>2</sub>CH<sub>2</sub>(CO<sub>2</sub>CH<sub>2</sub>CHMe<sub>2</sub>)CH<sub>2</sub>CO<sub>2</sub>CH<sub>2</sub>CHMe<sub>2</sub>, b<sub>2</sub> 124-8°, 1.1098, 1.4773, nontoxic; (PrO)<sub>2</sub>P<sub>2</sub>CH<sub>2</sub>(CO<sub>2</sub>Et)CH<sub>2</sub>CO<sub>2</sub>Et, b<sub>2</sub> 145°, 1.1700, 1.4890, 0.069; (PrO)<sub>2</sub>P<sub>2</sub>CH<sub>2</sub>(CO<sub>2</sub>CHMe<sub>2</sub>)CH<sub>2</sub>CO<sub>2</sub>CHMe<sub>2</sub>, b<sub>2</sub> 125-5°, 1.1146, 1.4785, nontoxic; (PrO)<sub>2</sub>P<sub>2</sub>CH<sub>2</sub>(CO<sub>2</sub>Bu)CH<sub>2</sub>CO<sub>2</sub>Bu, b<sub>2</sub> 143-5°, 1.0947, 1.4822, about 0.5; (iso-PrO)<sub>2</sub>P<sub>2</sub>CH<sub>2</sub>(CO<sub>2</sub>Et)CH<sub>2</sub>CO<sub>2</sub>Et, b<sub>2</sub> 151°, 1.0702, 1.5440, 0.25; (BuO)<sub>2</sub>P<sub>2</sub>CH<sub>2</sub>(CO<sub>2</sub>Et)CH<sub>2</sub>CO<sub>2</sub>Et, b<sub>2</sub> 125-8°, 1.1078, 1.4801, >0.5; (iso-BuO)<sub>2</sub>P<sub>2</sub>CH<sub>2</sub>(CO<sub>2</sub>Et)CH<sub>2</sub>CO<sub>2</sub>Et, b<sub>2</sub> 117-30°, 1.0042, 1.4855, >0.5. G. M. Kosolapoff.

Translation T 139 R, 16 June 57

MEL'NIKOV, N.N.

Use of labelled atoms for studying the stability of insecticide dusts containing organic thiophosphates. R. A. Gay, N. N. Mel'nikov, Ya. A. Mandelbaum, V. I. Chesnokova and ~~G. P. Shvartsova-Shirokova (Dokl. Akad. Nauk SSSR, 1954, 84, 729-732).~~  
The rate of loss of P from 1% dusts of diethyl p-nitrophenylthiophosphate containing  $^{32}\text{P}$  and/or  $^{33}\text{S}$  is greater, at the same temp., than that from ethyl pp'-dinitrodiphenylthiophosphate, and for both compounds increases with temp. (measurements at 15, 22, and 45°); it is also greater in the light than the dark. The decrease in toxicity runs parallel to the loss of P. The major part of the toxicity of these preparations will have vanished after 4 e<sup>-7</sup>s exposure on crops under normal conditions.

R. C. MURRAY.

*Mel'nikov, N. N.*

An application of the method of labeled atoms in the study of resistance of *Urygaster latroscriptus* to two organophosphorus insecticides and experimental study of their penetration into the plants. K. A. Gar, Yu. A. Mandozhan, N. N. Mel'nikov, E. D. Shvetsova-Shilovskaya, and V. I. Chirnetsov. *Doklady Akad. Nauk S.S.S.R.* 94, 1180-82 (1954).

<sup>32</sup>P-labeled specimens of (EtO)<sub>2</sub>P(O)C<sub>6</sub>H<sub>4</sub>NO<sub>2</sub> and EtOP(O)(OC<sub>2</sub>H<sub>5</sub>)NO<sub>2</sub> were used in 1% dusts which were applied to male and female specimens of the insects. Females were generally more resistant to both insecticides than the males. A direct relation was found between the amt. of P which penetrates the insect body and the degree of poisoning, within each capt. group. Death occurs with lower level of the di-Et deriv. than mono-Et deriv., but this is caused not by a mere difference of diffusion, since in dead specimens the difference in permeability disappears between females and males. Chrysanthemum plants were allowed to absorb through the roots aq. emulsions of the di-Et deriv. (0.05-0.2%) and the penetration to the leaves was studied radioactively. A spraying with even 0.2% emulsion failed to give complete control of *Aulacothrips betargonii*, although the amt. of the insecticide which penetrated the plant mass reached 0.003% of the green mass at room temp. This corresponds to 20-30 mg./kg. At lower temp. when this value reached 60 mg./kg. a considerable degree of control was attained and the insects contained up to 22 mg./kg. of the di-Et deriv. The penetration into chrysanthemum was substantially like that found in beet. However, on cabbage cultures no control was achieved by this method against *Drepanocoryne brassicae*, although withering of leaves was observed at 0.05% concn. of the emulsion, or higher. In cabbage and chrysanthemum capt., considerable hydrolysis of the insecticide took place and after 30 days only the hydrolysis products remained; this process is accelerated by sunlight. Dusting with 1% dust on shaded kidney beans showed 42% hydrolysis after 16 days; in sunlight almost all was hydrolyzed in 4 days. On wheat the process takes but 3 days. Thus parathion is not truly a systemic insecticide, owing to its poor penetration and stability in the plant.

G. M. Kosolapoff

MEL'NIROV, N. N.

USSR/Chemistry

Card : 1/1

Authors : Mandel'baum, Ya. A., Lomskina, V. I. and Mel'nikov, N. N.

Title : Synthesis of dimethyl-4-nitrophenylthiophosphate and trimethylthiophosphate marked with radioactive phosphorus ( $P^{32}$ ).

Periodical : Dokl. AN SSSR, 96, Ed. 6, 1173 - 1174, June 1954

Abstract : In connection with the investigation of new phosphor-containing insecticides the authors synthesized dimethyl-4-nitrophenylthiophosphate and trimethylthiophosphate marked with radioactive phosphorus. The synthesis of above compounds was realized on the basis of the following reactions:  $PSCl_2 + 2CH_3OH + 2NaOH \rightarrow (CH_3O)_2 PSCl + 2NaCl + 2H_2O$ ;  $(CH_3O)_2 PSCl + HOC_6H_4NO_2 + Na_2CO_3 \rightarrow (CH_3O)_2 PSOC_6H_4NO_2 + 2NaHCO_3 + NaCl$ . Two references.

Institution : Scientific Institute for Development of Fertilizers and Insecticides

Presented by : Academician S. I. Vol'fkovich, March 17, 1954

MEL'NIKOV, N. N.

USSR/Chemistry

Card 1/1

Authors : Bokarev, K. S., and Mel'nikov, N. N.

Title : Synthesis of 4-iodophenoxyacetic acid, marked  $J^{131}$

Periodical : Dokl. AN SSSR, 97, Ed. 2, 255 - 256, July 1954

Abstract : The method for the derivation of 4-iodophenoxyacetic acid marked with radioactive  $J^{131}$  is presented. The selection of a special method for the derivation of the compound should take into consideration the comparatively small period of semi-decomposition of the radioactive iodine. The introduction of the radioactive iodine takes place in the last phase of the process and consumes only several hours. Three references.

Institution : Acad. of Sc. USSR, The K. A. Timiryazev Institute of Plant Physiology

Presented by : Academician S. I. Volkovich, March 22, 1954

MEL'NIKOV, N. N.

USSR/Chemistry

Card : 1/1

Authors : Baskakov, Yu. A. and Mel'nikov, N. N.

Title : Synthesis of alpha-naphthylacetic acid and its methyl ether, marked  $C^{14}$  in the carboxyl group

Periodical : Dokl. AN SSSR, 97, Ed. 3, 453 - 454, July 21, 1954

Abstract : The derivation of methyl ether of alpha-naphthylacetic acid, and marked  $C^{14}$  in carboxyl, through the esterification of the acid with methyl alcohol, in the presence of sulfuric acid, is described. Formulas for the synthesis of such compounds, are given. Methyl ether of alpha-naphthylacetic acid is a plant growth stimulus and its various applications are outlined. Six USSR references.

Institution : Acad. of Sc. USSR., The K. A. Timiryazev Institute of Plant Physiology

Presented by : Academician, S. I. Vol'fkovich, March 22, 1954

Mel'nikov, N.N.

Preparation of dialkyl chlorophosphates. N. N. Mel'nikov and Ya. A. Mandelbaum (Sverdlov Sci. Research Inst. Fertilizers and Insectofungicides, Moscow). *Khim. i Primenenie Fosfororgan. Soedinenii, Akad. Nauk S.S.S.R. Trudy 1-oi Konferentsii*, 1955, 185-93 (Pub. 1957).—Prepn of  $(EtO)_2PSCl$  from  $PCl_3$  and  $EtOH$  gives 50-6% yields if the reaction is run in the presence of an org. base, best in  $(CH_3)_3Cl$ .—Use of  $C_6H_6$  as the solvent lowers the yield, while  $CCl_4$  introduces unstated impurities, besides yielding up to 30%  $EtOPSCl_2$ . Presence of  $H_2O$  or excess  $EtOH$  lowers the yield owing to formation of the pyro ester.  $Et_3N$  or  $PhNMe_2$  as the acid removing agents are unsatisfactory owing to formation of much  $EtOPSCl_2$ . Pyridine or pyridine bases are most satisfactory. In a typical run 255 g.  $PCl_3$  in 0.6 vols.  $(C_6H_5)_3Cl$  was treated at  $-5^\circ$  with 2.17 moles dry  $EtOH$  in 2.44 moles pyridine; after 1-2 hrs. stirring at  $10-20^\circ$ , the mixt. was filtered and the filtrate washed with  $H_2O$ , dried, and distd., yielding 50-5%  $(EtO)_2PSCl$  and 7-10%  $EtOPSCl_2$ . With crude picoline bases the reaction can be run without a solvent yielding 43-6% desired chloride. Deficiency of the org. base results in evolution of much  $EtCl$ . Yields of 60-8% are attained also by adding 1 mole dry  $EtOH$  to 1 mole  $PCl_3$  (temp. rise to  $30-40^\circ$ ), followed by addn. of 0.6 mole  $EtOH$  (temp. rise to  $45-7^\circ$ ) with percolation of the mixt. with  $N$  3 hrs. at this temp.; when the mixt. is cooled under  $15^\circ$  and treated slowly with  $NaOH$  in abs.  $EtOH$  (0.1 g.  $NaOH$  in 55 ml.  $EtOH$  per 0.1 mole  $PCl_3$ ), kept 1 hr. at  $20^\circ$ , dilld. with  $H_2O$ , dried, and distd., there is formed 60-65%  $(EtO)_2PSCl$ . Solid  $NaOH$  or its aq. soln. can be used also. The product contains 3.6%  $EtOPSCl_2$ . A yield of 80% is attainable as follows: to  $Mg$  shavings is added 0.05 g. iodine and a little  $CCl_4$  after which enough abs.  $EtOH$  (not over 0.3% moisture) is added to

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MELNIKOV, N. N., MITCHELL, D. A. A.

cover the Mg and the reaction is allowed to commence; the addn. of EtOH is continued, keeping the mixt. boiling until all the Mg is dissolved (4.5-5 moles/g. atom Mg is needed, since the Mg ethoxide contg. 2 moles EtOH of crystal. gives the best results). A suspension of 1.1 mole  $(EtO)_2Mg$  in abs. EtOH is then added to 1 mole  $PSCl_2$  below  $35^\circ$  over 50 min.; after stirring 1-2 hrs. at  $45-50^\circ$ , the mixt. is cooled and dild. with  $H_2O$  (any ppt. of  $Mg(OH)_2$  is dissolved in  $HCl$ , the sepd. oil being sepd., dried, and distd. The product contains in crude state 2-4%  $(EtO)_2PSCl_2$  and up to 5%  $(EtO)_3PS$ . If  $PSCl_2$  contains over 2-3%  $PCl_5$ , the yield is severely reduced. A simpler procedure: to 340 g.  $PSCl_2$ , was added in 15-30 min. 150 ml. abs. EtOH (temp. rise to  $35-40^\circ$ ) after which dry N is percolated through the mixt. at  $45-7^\circ$  3-3.5 hrs. until  $HCl$  evolution stops, then the percolation is continued 40 min. to remove residual  $HCl$  at  $10-15^\circ$ ; the residue is treated over 40 min. with  $(EtO)_2Mg$  prepd. from 31 g. Mg and 650 ml. abs. EtOH; the addn. is made at  $20-25^\circ$ , after which the mixt. is stirred 60-80 min. at  $50^\circ$ , cooled and quenched in  $H_2O$ ; the yield of pure  $(EtO)_2PSCl_2$  is 72%, with 5%  $(EtO)_3PS$ ; and 1.5%  $(EtO)_3PS$  also being formed. Use of  $(EtO)_2Al$  in this reaction gives but 40-50% yields and leads to some loss of S by the P component. For prepn. of  $(MeO)_2PSCl_2$  the aq. NaOH method (above) is best. The calcd. amt. of MeOH is added to  $PSCl_2$  at  $-8^\circ$ , followed by NaOH (in aq. soln., in MeOH, or as solid powder) below  $0^\circ$ ; 10% excess NaOH is recommended. After keeping at  $20^\circ$  until the reaction is complete (tested by siphoning a sample and isolating the product), the mixt. is dild. with  $H_2O$  and worked up as above. — Yields are 70-80%; a soln. of NaOH in 60% aq. MeOH can be used. It is stated that  $(MeO)(EtO)PSCl_2$  can be made similarly in 85-91% yield.

G. M. Kosolapoff

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MEL'NIKOV, N.H.

MEL'NIKOV, N.H.

Utilizing organosynthetic industrial products in agriculture.  
Soob.o nauch.rab.chl.VKHO no.2:17-32 '55. (MIRA 10:10)  
(Agricultural chemicals)

MeL'NIKOV N.N.

Structure and physiological activity of alkyl and aryl  
 phenoxycarboxylic acids and their derivatives. N. N.  
 MeL'nikov, R. Kh. Turfakova, Izv. Akad. Nauk SSSR, Ser. Khim. Nauk, 1967, No. 1, p. 1155.  
 (1) is presented which have been tested for bacterial and  
 growth-stimulating properties. Halogenation of 1 causes a  
 small increase in activity of the compound. Introduction of a  
 2nd halogen exerts a neg. effect. A trihalo deriv. with the  
 3rd halogen in the 4' position has greater activity than the  
 dihalo deriv. Having transition from the chloro to the  
 bromo deriv., the activity of the compound decreases. 2-  
 Phenyl-4-halo-1 has greater activity than its isomer 4-phenyl-  
 2-halo-1. A similar analogy is observed with the 3-halo-  
 2-phenyl-6-halo-1 has less activity than acids contg. the  
 halogen in position 4. Analogues of the acids have less ac-  
 tivity than the acids. The activity of the amides is still  
 lower. Halogen substituents of naphthoalkylcarboxylic  
 acids have less activity than those contg. halogens.  
 Alkyl-1 which contain hydrocarbon radicals in position 4 and  
 their halogen derivs. are considerably less active than  
 compounds contg. hydrocarbon radicals in position 2; 4-  
 benzyl-1 and its halogen derivs. are comparatively less ac-  
 tive than the 2-Me-1 and its halogen derivs. Acetylphen-  
 oxyacetic acids have extremely little activity. This group  
 of compounds possesses lower activity which is apparently  
 related to their high mol. wt. and their slow rate of diffu-  
 sion. Wurz. H. Pharmazie

M. N. Kh.

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MEL, NIKOV, N N

FD-3375

USSR/Chemistry - Defoliants

Card 1/1 Pub. 50 - 19/20

Authors : Baskakov, Yu. A., Mel'nikov, N. N.

Title : Chemical agents for the defoliation and dessication of plants (Foreign developments)

Periodical : Khim. prom. No 7, 435-443, Oct-Nov 1955

Abstract : Review on the basis of foreign publications the subject of defoliants and plant dessicants. Three USSR references, all since 1940; 107 foreign references.

Institution : --

MEL'NIKOV, N. N.

AID P - 3172

Subject : USSR/Chemistry

Card 1/1 Pub. 119 - 7/8

Authors : Vol'fkovich, S. I. and V. K. Kuskov

Title : N. N. Mel'nikov, Yu. A. Baskakov, Khimiya gerbitsidov i stimulyatorov rosta rasteniy (Chemistry of herbicides and growth-promoting substances for plants) Moscow, 1954. (Book Review)

Periodical : Usp. khim., 24, 5, 635-636, 1955

Abstract : Critical review.

Institution : None

Submitted : No date

MEI'NIKOV N.N.

**Organic Insecticides. XVII. Synthesis of amides and hydrazides of diazoxyphosphoric acids.** N. N. Mei'nikov and A. G. Zen'kovich. *Zhur. Obshchei Khim.* 1955, 29, 2573-2575. *J. Gen. Chem. U.S.S.R.* 25, 763-5(1955) (Engl. translation); *C.A.* 48, 986g, 10649c. — While compds. (RO)<sub>2</sub>PSNH<sub>2</sub> show systemic insecticidal activity, this lies below that of octamethylpyrophosphoramide. Similar and weak activity was found in (MeO)<sub>2</sub>PSNMe<sub>2</sub> and (MeO)<sub>2</sub>PSNEt<sub>3</sub>. Weak contact insecticidal activity is characteristic of the various amides shown below. The compounds were prepd. from (RO)<sub>2</sub>PSCl which, in unstated solvent, were treated with the desired amine, NH<sub>3</sub>, or N<sub>2</sub>H<sub>4</sub>. The amine HCl salt was filtered and the soln. distd. *in vacuo*; butan removal of amine HCl salt was obtained in some cases by washing the soln. with H<sub>2</sub>O after filtration. Thus were prepd. the following (b.p., d<sub>4</sub>, n<sub>D</sub><sup>20</sup> shown, resp.): (MeO)<sub>2</sub>PSNH<sub>2</sub>, b<sub>p</sub> 105-8°, 1.2648, 1.4082; (EtO)<sub>2</sub>PSNH<sub>2</sub>, b<sub>p</sub> 113-15°, 1.1456, 1.4328; (PrO)<sub>2</sub>PSNH<sub>2</sub>, b<sub>p</sub> 5-62°, 1.0833, 1.4798; (iso-PrO)<sub>2</sub>PSNH<sub>2</sub>, b<sub>p</sub> 82°, 1.0662, 1.4670; (BuO)<sub>2</sub>PSNH<sub>2</sub>, b<sub>p</sub> 83-8°, 1.0107, 1.4715; (MeO)<sub>2</sub>PSNMe<sub>2</sub>, b<sub>p</sub> 77-8°, 1.1712, 1.4766; (EtO)<sub>2</sub>PSNMe<sub>2</sub>, b<sub>p</sub> 97-100°, 1.0622, 1.4635; (PrO)<sub>2</sub>PSNMe<sub>2</sub>, b<sub>p</sub> 79-82°, 1.0232, 1.4646; (iso-PrO)<sub>2</sub>PSNMe<sub>2</sub>, b<sub>p</sub> 71°, 1.0070, 1.4583;

(BuO)<sub>2</sub>PSNMe<sub>2</sub>, b<sub>p</sub> 121°, 0.9958, 1.4623; (MeO)<sub>2</sub>PSNEt<sub>3</sub>, b<sub>p</sub> 88-91°, 1.0406, 1.4780; (PrO)<sub>2</sub>PSNEt<sub>3</sub>, b<sub>p</sub> 81-3°, 1.0029, 1.4628; (iso-PrO)<sub>2</sub>PSNEt<sub>3</sub>, b<sub>p</sub> 100°, 1.0210, 1.4620; (BuO)<sub>2</sub>PSNEt<sub>3</sub>, b<sub>p</sub> 91-3°, 0.9783, 1.4610; (MeO)<sub>2</sub>PSNEtPh, b<sub>p</sub> 87-88°, 1.1901, 1.5535; (EtO)<sub>2</sub>PSNEtPh, b<sub>p</sub> 81-107°, 1.1573, 1.6499; (PrO)<sub>2</sub>PSNEtPh, b<sub>p</sub> 101-5°, 1.0904, 1.6272; (iso-PrO)<sub>2</sub>PSNEtPh, m. 31°; (BuO)<sub>2</sub>PSNEtPh, b<sub>p</sub> 115°, 1.0697, 1.5245; (MeO)<sub>2</sub>PSN(CH<sub>2</sub>CH<sub>2</sub>OH)<sub>2</sub>, b<sub>p</sub> 130°, 1.2470, 1.5090; (EtO)<sub>2</sub>PSN(CH<sub>2</sub>CH<sub>2</sub>OH)<sub>2</sub>, b<sub>p</sub> 143-6°, 1.1742, 1.4945; (PrO)<sub>2</sub>PSN(CH<sub>2</sub>CH<sub>2</sub>OH)<sub>2</sub>, b<sub>p</sub> 149-51°, 1.1313, 1.4900; (iso-PrO)<sub>2</sub>PSN(CH<sub>2</sub>CH<sub>2</sub>OH)<sub>2</sub>, b<sub>p</sub> 103-10°, 1.1282, 1.4820; (BuO)<sub>2</sub>PSN(CH<sub>2</sub>CH<sub>2</sub>OH)<sub>2</sub>, b<sub>p</sub> 170-65°, 1.1759, 1.4350; (MeO)<sub>2</sub>PSNHNH<sub>2</sub>, m. 103-9°; (EtO)<sub>2</sub>PSNHNH<sub>2</sub>, m. 83-6°; (PrO)<sub>2</sub>PSNHNH<sub>2</sub>, b<sub>p</sub> 76-80°, 1.1031, 1.4896; (iso-PrO)<sub>2</sub>PSNHNH<sub>2</sub>, m. 123-9°; (BuO)<sub>2</sub>PSNHNH<sub>2</sub>, b<sub>p</sub> 110-15°, 1.0593, 1.5410; (MeO)<sub>2</sub>PSNHNHPh, m. 85-5°; (EtO)<sub>2</sub>PSNHNHPh, m. 92-5°; (PrO)<sub>2</sub>PSNHNHPh, m. 87-91°; (iso-PrO)<sub>2</sub>PSNHNHPh, m. 42-5°; (BuO)<sub>2</sub>PSNHNHPh, m. 19-21°.

G. M. Kasolanzoff

MEL'NIKOV, N.N.; ZEN'KEVICH, A.G.

Organic insecticide-fungicides. Part 17. Synthesis of amides and hydrazides of dialkoxythiophosphoric acids. Zhur.ob.khim. 25 no.4: 828-831 Ap '55. (MLRA 8:7)

1. Nauchnyy institut po udobreniyam i insektofungitsidam.  
(Insecticides) (Fungicides) (Phosphorus organic compounds)

MEL'NIKOV, N. N.

USSR/Organic Chemistry - Naturally Occurring Substances and Their Synthetic  
Analogues, E.

Abst Journal: Referat Zhur - Khimiya, No 19, 1956, 61617

Author: Bokarev, K. S., Mel'nikov, N. N.

Institution: None

Title: Synthesis of Some New Glucose Esters

Original

Periodical: Zh. obshch. khimii, 1955, 25, No 12, 2242-2245

Abstract: For the purpose of a study of the mechanism of the action of derivatives of phenoxyacetic acid upon plants there have been synthesized 2,3,4,6-tetraacetyl-1-(2',4'-dichlorophenoxyacetyl)- (I) and 2,3,4,6-tetraacetyl-1-(2',4',5'-trichlorophenoxyacetyl)-glucose (II). On disacetylation of I and II with a solution of  $\text{NH}_3$  in  $\text{CH}_3\text{OH}$  there are formed quantitatively the corresponding chlorophenoxyacetamides. Formation of similar glucose esters in plants is improbable. By boiling of a mixture of 30 g Ag-salt of 2,4-dichlorophenoxyacetic acid, 41.1 g acetobromoglucose and 150 ml

Card 1/2

USSR/Organic Chemistry - Naturally Occurring Substances and Their Synthetic  
Analogues, E-3

Abst Journal: Referat Zhur - Khimiya, No 19, 1956, 61617

Abstract:  $C_6H_6$  (10 hours) I has been obtained with almost a quantitative yield, MP 125-126° (from  $CH_3OH$ ),  $[\alpha]^{20}_D + 7.4^\circ$  (dioxane). Analogously has been prepared II, yield 78.3%, MP 170-171° (from  $CH_3OH$ ),  $[\alpha]^{20}_D + 8.3^\circ$  (dioxane). I and II have  $\beta$ -configuration since they are saponified by emulsion.

Card 1/2

MEL'NIKOV, N.N.

Synthesis of N-2,4,5-trichlorophenoxyacetyl amino acids.  
K. S. Bokarev and N. N. Mel'nikov. *J. Gen. Chem. U.S.S.R.*  
S.R. 25, 2385-7 (1955) (English translation). See C.A. 50,  
9381g. R.M.R.

*Chem*

2

MELNIKOVA - N.N.

Synthesis of some N-2:4:5-trichlorophenoxyacetamido acids  
 R. S. Bokarek and N. N. Melnikova (Zh. obshch. Khim. 1955, 24, 2489).  
 2:4:5-Trichlorophenoxyacetyl chloride, with  
 suitable amino acids in the presence of alkali or pyridine,  
 compounds showed various physiological activities when applied to  
 plants but were generally found to stimulate their growth. Syn-  
 thesis of N-2:4:5-trichlorophenoxyacetamido acids by a similar  
 method from 2:4:5-trichlorophenoxyacetyl chloride yielded more  
 active compounds. Complete data and classification are tabulated.  
 A. L. R.

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*Met. fiziologii. Trudy, Acad. Nauk SSSR.*

MEL'NIKOV, N. N.

13

Preparation of 1-naphthylacetic acid by the Willgerodt reaction. Yu. A. Baskakov and N. N. Mel'nikov, *Zhur. Priklad. Khim.* 28, 1010-18 (1955) ~~1955~~ 1956, 380 g. ( $\text{CH}_2\text{Cl}$ ), 183.6 g.  $\text{Ac}_2\text{O}$ , and 205 g.  $\text{C}_{10}\text{H}_8$ , kept below 2° 2 hrs.; gave 92-3.5% 1-naphthyl *Ac ketone*,  $b_p$  143-5°,  $b_m$  154-7°; the product is essentially free of the 2-isomer. The ketone (20 g.), 20 ml. EtOH, and 20 ml.  $(\text{NH}_4)_2\text{S}_8$  heated in sealed capsule 4-5 hrs. at 155-65° (60-60 atm.) gave after filtration and treatment of the solid residue with  $\text{CCl}_4$ , 81-3.5% 1-naphthylacetamide,  $m.$  190-2°. This refluxed 6 hrs. with aq. NaOH (6 g. NaOH per 54 ml.  $\text{H}_2\text{O}$ ) gave 68% 1-naphthylacetic acid,  $m.$  131°.

CH

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MA

MA

MEL'NIKOV, N. N.

USSR/Chemistry - Insecticides

Card 1/1 Pub. 22 - 20/50

Authors : Mandel'baum, Ya. A.; Vladimirova, I. L.; and Mel'nikov, N. N.

Title : Synthesis of diethyl-4-nitrophenylthiophosphate and ethyl-4,4'-dinitrophenylthiophosphate marked with radioactive P<sup>32</sup> and S<sup>35</sup>

Periodical : Dok. AN SSSR 100/1. 77-79. Jan 1. 1955

Abstract : The synthesis of insecticides containing phosphor (diethyl-4-nitrophenylthiophosphate and ethyl-4,4'-dinitrodiphenylthiophosphate), is described. The methods employed in the synthesis of the insecticides were first tested on inactive substances. In selecting the proper synthesis method it was necessary to take into consideration the comparatively short period of P<sup>32</sup> semi-decomposition. The results obtained during the synthesis with marked radioactive P<sup>32</sup> and S<sup>35</sup> are listed. Two USSR references (-).

Institution : The Ya. V. Semoilov Scientific Institute on Matters of Fertilizers and Insecticides

Presented by: Academician S. I. Vol'fkovich, March 17, 1954

MEL'NIKOV, N.N.

Reaction between hexachlorocyclopentadiene and some unsaturated compounds. L. G. Vol'ison, N. N. Mel'nikov, A. F. Platé, Yu. N. Sapozhkov and G. S. Tatts (*Dokl. Akad. Nauk SSSR*, 1955, 105, 1252-1255). Hexachlorocyclopentadiene is condensed with a series of unsaturated cyclic compounds, to give the following products (figures in parentheses refer, respectively, to yield and to toxicity to insects, taking chlordane as 100): 4:5:6:7:8:8-hexachloro-4:7-endomethylene-1:2:3a:4:7:7a: hexahydroindene (50%; 11), m.p. 162-185° (from cyclopentene; 6 hr. at 120-125°); 4:5:6:7:8:8-hexachloro-4:7-endomethylene-1-methyl-1:2:3a:4:7:7a-hexahydroindene (yield not given; 30), m.p. 51-53° (from 2-methylcyclopentene; 5 hr. at 120°); 1:2:3:3:3a:7a-hexachloro-4:7-endoxy-7-methyl-3a:4:7:7a-tetrahydroindene (yield small; 30), m.p. 175-176° (from 2-methylfuran; 7 hr. at 70°); 1:2:3:4:9:9-hexachloro-1:4-endo-methylene-1:4:4a:5:6:7:8:8a-octahydronaphthalene (33%; 50) (from cyclohexene; 30 hr. at 115°); 1:2:3:4:10:10-hexachloro-1:4:5:8-diendomethylene-1:4:4a:5:6:8a-hexahydronaphthalene (aldrin; yield not given; 200) (from bicyclo-2:2:1-hepta-2:5-diene; 25 hr. at 100°); 1:2:3:4:10:10-hexachloro-1:4:5:8-diendomethylene- (76%; 100) (from bicyclo-2:2:1-hept-2-ene; 6-5 hr. at 180°); 1:2:3:4:10:10-hexachloro-1:4:5:8-diendomethylene-6-methyl- (46%; 9), m.p. 55-58° (from 5-methylbicyclo-2:2:1-hept-2-ene; 11 hr. at 150°); 1:2:3:4:10:10-hexachloro-1:4:5:8-diendomethylene-6-ethyl- (yield not given; 14); m.p. 56-58° (from 5-ethylbicyclo-2:2:1-hept-2-ene; 13 hr. at 150°); 1:2:3:4:10:10-hexachloro-1:4:5:8-diendomethylene-6-n-pentyl- (64%; toxicity not given), b.p. 188°/0.1 mm.,  $n_D^{20}$  1.5403,  $d_4^{20}$  1.3604 (from 5-n-pentylbicyclo-2:2:1-hept-2-ene; 27 hr. at 120-130°), and 1:2:3:4:10:10-hexachloro-1:4:5:8-diendomethylene-6-acetoxy-1:4:5a:5:6:7:8:8a-octahydronaphthalene (33%; 20) (from 5-acetoxycyclo-2:2:1-hept-2-ene; 23 hr. at 95°); and 1:2:3:4:7:7-hexachloro-5-phenyl-bicyclo-2:2:1-hept-2-ene (87%; 30) (from styrene; 3 hr. at 165-170°). The products are resistant to weathering, and do not lose Cl after boiling with N-NaOH in methanol for 1 hr.

R. Truscov.

VOL'FSON, L.G.; MEL'NIKOV, N.M.

Effect of fillers on the stability of DDT dusts. [Trudy] NIUIF  
no.156:64-73 '55. (MLRA 9:10)

(DDT (Insecticide))

W. J. ... and ...

"Preparation of ..."  
... presented at the First Conference ...  
... Dec 14

3 : B-3, 84, 811

MEL'NIKOV, N. N.

"Exploratory Work by Insectofungicide Institute Among  
Organophosphorous Insecticides"  
paper presented at Nn First Conference on Phosphorous Compounds, Kazan,  
8-10 Dec 56

SO: B-3,084,841

ML LNIKOV, N.N.

~~Methyl ethyl 2-ethylmercaptosethylthiophosphate. N.N.  
 Molnisev, Ya. A. Mandel'shtam, V. I. Lomakina, and  
 P. V. Zhov. U.S.S.R. 104,225, Nov. 25, 1958. A mixt.  
 of MeEtP(S)Cl and HOCH<sub>2</sub>CH<sub>2</sub>SEt with NaOH gives  
 MeEtP(S)OCH<sub>2</sub>CH<sub>2</sub>SEt, used as an insecticide.~~

M. Hosh

gmb

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4

RODIONOV, V.M., akademik, redaktor [deceased]; KAZANSKIY, B.A., akademik, redaktor; KNUBYANETS, I.L., akademik, redaktor; SHEMYAKIN, M.M., redaktor; MEL'NIKOV, N.H., professor, redaktor; TAYTS, S.Z., redaktor; SHEMASTINA, Ye.V., redaktor; KORNEYEVA, V.I., tekhnicheskiy redaktor

[Reactions and methods of analysis of organic compounds] Reaktsii i metody issledovaniia organicheskikh soedinenii. Moskva, Gos. nauchno-tekhn. izd-vo khim. lit-ry. Vol.4. 1956. 319 p. (MLBA 9:7)

1. Chlen-korrespondent AN SSSR (for Shemyakin)  
(Chemical reactions) (Isomers and isomerization)

POPOV, Petr Vasil'yevich; YEGOROV, N.G.,redaktor; MEL'NIKOV, N.N.,  
professor, redaktor; LUR'YE, H.S.,tehnicheskij redaktor

[Reference manual of toxic chemicals] Spravochnik po iadokhimitam.  
Pod red. N.N. Mel'nikova. Moskva, Gos. nauchno-tekhn. izd-vo khim.  
lit-ry, 1956. 623 p. (MLRA 10:5)  
(Agricultural chemicals)

MELNIKOV, N. N.

USSR/Special and General Zoology - Insects.

0-3

Abs Jour : Referat Zhur - Biologiya, No 16, 1957, 69838

Author : Melnikov, N.N.

Inst :

Title : The Chemical Means of Plant Protection and Their Application.

Orig Pub : Khim. nauka i proo-st'. , 1956, 1, No 2, 160-173

Abstract : The tremendous growth in the production of chemicals for the protection of plants, and of weed-killers is noted. The world production of chloro- and organophosphorus preparations may consist of hundreds and thousands of tons. In USA the production of basic weed-killers in 1950-1954 was about 275 thousand tons a year. The assortment of preparations is constantly increasing. And so, in the USA in 1953 about 5100 preparations, and in 1955 over 6000, and in Germany 1000 substances were produced, but the active radical of these preparations may be limited

Card 1/2

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... given. ... and/or compounds; derivatives of dithiocarbamic acid, quinones, 3-n-chlorophenyl-5-methylrhodamine; fungicides with systemic effect; antibiotics; mordants - organic Hg-compounds, hexachlorobenzol, 2,4,5-trichlorophenolates of Cu. As present killers the following should be noted:

Thallium sulfate and Zinc phosphide; squill, (glucoside, skilliraside); a-naphthylthiourea, and n-chlorophenyldiazothiourea; salts of fluoro-acetic acid and fluoro-acetamid; anticoagulants- production of coumarine et al.

A special chapter is developed to herbicides, defoliants and dessicants.

Card 2/2

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MEL'NIKOV, N.N.

Conference on defoliation. Khim. prom. no.3:184 Ap-My '56.  
(MLRA 9:10)

(Agricultural chemicals)

BASKAKOV, Yu.A.; MEL'NIKOV, N.N.

On transformations of growth stimulators in plants. (1-alkylindolyl-3-acetic acids) [with English summary in insert]. Fiziol.rast. 3 no.3:208-213 My-Je '56. (MIRA 9:9)

1. Institut fiziologii rasteniy imeni K.A.Timiryazeva i Institut biokhimii imeni A.N.Bakha Akademii nauk SSSR, Moskva.  
(Growth promoting substances) (Acetic acid)

*Melnikov, V. A.*  
 Synthesis of some derivatives of 3,6-endoxohexahydrophthalic acid. N. N. Melnikov and V. A. Krut', *Zhur. Obshch. Khim.* 26, 213-218, *F. Gen. Chem. U.S.S.R.* 26, 221-22 (1956) (Engl. translation).—Adducts of maleic anhydride with furan and sylvan were hydrogenated over Pd on metallic Ni (cf. Ginzberg and Ivanov, *C.A.* 25, 4173) in Me<sub>2</sub>CO or H<sub>2</sub>O soln. Thus were prepd: 3,6-endoxohexahydrophthalic anhydride (I), m. 110-17°, and the 3-methyl deriv., m. 105-6°. The hydrogenations run in aq. Na<sub>2</sub>CO<sub>3</sub> gave the free acids which were converted to the anhydrides by treatment with AcCl. The anhydrides were treated with amines in dioxane soln., best run with cooling, yielding the following *N-R*-substituted 3,6-endoxohexahydrophthalamic acids (*R* shown): *N,N*-diethyl, 84.6%, m. 132-3°; *N-Ph*, 100%, decomp. 161°; *N-I-C<sub>6</sub>H<sub>5</sub>*, 83.3%, m. 164°. Heating these briefly gave loss of H<sub>2</sub>O and formed 100% *N*-phenylimide of 3,6-endoxohexahydrophthalic acid, m. 163-4°; the 3-Me deriv., m. 210-20°. Similarly were prepd: *N-R*-substituted 3-methyl-3,6-endoxohexahydrophthalamic acids: *N,N*-dimethyl, 100%, m. 122-3°; *N,N*-diethyl, 90.2%, m. 129-30°; *N-Ph*, 85.2%, m. 144-5°; *N-I-C<sub>6</sub>H<sub>5</sub>*, 77.8%, m. 143.5-4°; and the *N*-phenylimide of 3-methyl-3,6-endoxohexahydrophthalic acid, m. 179.5-90°. The anhydrides treated with alcs. in the presence of a catalytic amt. of PhSO<sub>2</sub>Li, using CH<sub>2</sub>ClCH<sub>2</sub>Cl as the azeotrope with H<sub>2</sub>O, gave the following esters: 3,6-endoxohexahydrophthalic acid di-*Pr* ester, 73.5%, m. 55-8°; di-*iso-Pr* ester, 69%, m. 103-4°; di-*Bu* ester, 73.6%, m. 61-2°; di-*iso-Am* ester, m. 80-7°, 89%; 3-methyl-3,6-endoxohexahydrophthalic acid di-*Pr* ester, 77.5%, m. 61-2°; di-*iso-Pr* ester, 81.5%, m. 78.5-7.6°; di-*Bu* ester, 83.5%, m. 49-4°. Chlorination of I in CHCl<sub>3</sub> at 0-5° 3 hrs. gave 70.3% 3,6-endoxo-4,5-dichlorohexahydrophthalic anhydride, m. 161-2°; the 3-methyl deriv., prepd. similarly, m. 151-2°. These appear to be *exo-cis* isomers; the last product was accompanied by an isomer, m. 107°, apparently the *gem*-metric isomer of the high-melting form. G. M. K.

2

WOOD

MEL'NIKOV, N. N.

Научный институт по  
исследованиям насекомых

✓ Organic insectofungicides. XVIII. New method of preparation of esters of chloro- and dichlorothiophosphoric acids. Z. M. Bakanova, Ya. A. Mandel'baum, N. N. Mel'nikov, and E. I. Svetsitskiĭ. *Zhur. Obshchei Khim.* 26, 494-5 (1956); cf. *C.A.* 50, 2415d.—Refluxing fine Al wire in 2-4 fold excess of abs. EtOH in the presence of 0.1 g. Hg(OAc)<sub>2</sub> and a little iodine for activation of Al, until all Al goes into soln. results in a rapid prepn. of Al(OEt)<sub>3</sub>. With an equimolar amt. of EtOH, C<sub>2</sub>H<sub>5</sub> is used as a diluent and the reaction is much slower. To 17 g. PSCl<sub>2</sub> there was added with cooling a soln. from 0.5 g. Al and 3 g. EtOH in 8 ml. C<sub>2</sub>H<sub>5</sub>; after 3 hrs. at 50° the mixt. was washed with ice-H<sub>2</sub>O acidified with HCl, dried, and distd., yielding 40% EtOPSCl<sub>2</sub>, b<sub>25</sub> 68°, d<sub>4</sub> 1.3968, n<sub>D</sub><sup>20</sup> 1.5030. To 34 g. PSCl<sub>2</sub> was added with cooling a soln. from 2 g. Al and 25 ml. EtOH; after 2 hrs. at 50-60° the cooled mixt. was washed with cold H<sub>2</sub>O acidified with HCl, yielding 42% (EtO)<sub>2</sub>PSCl, b<sub>25</sub> 98-8°, d<sub>4</sub> 1.2015, n<sub>D</sub><sup>20</sup> 1.4670. XIX. Synthesis of mixed esters of dithiophosphoric acid containing an amide group in the aliphatic ester radical. K. D. Shvetsova-Shilovskaya, N. N. Mel'nikov, and N. I. Marten'yukova. *Ibid.* 496-8.—Appropriate aldehydes and esters of carbamic acid were mixed and treated with (RO)<sub>2</sub>PS<sub>2</sub>H; after standing 1-3 days at room temp. the products were extrd. with C<sub>2</sub>H<sub>5</sub>, washed with H<sub>2</sub>O, dried and distd. No other details are given. Thus were prepd.: (RO)<sub>2</sub>P(S)SCH<sub>2</sub>NR'CO<sub>2</sub>R'' (R, R', R'', % yield, b. p., d<sub>4</sub>, and n<sub>D</sub><sup>20</sup> given): Me, H, Et, 30.3, b<sub>25</sub> 107-10°, 1.3498, 1.5591; Et, H, Et, 42, b<sub>25</sub> 64-8°, 1.1904, 1.4990; Pr, H, Et, 60.7, b<sub>25</sub> 82°, 1.0890, 1.4612; iso-Bu, H, Et, 46.6, b<sub>25</sub> 122-4°, — (m. 22°); Et, Me, Et, 20.8, b<sub>25</sub> 107-14°, 1.1814, 1.5041; Bu, Me, Et, 53.7, b<sub>25</sub> 145-52°, 1.0675, 1.4870; iso-Bu, Me, Et, 57.5, b<sub>25</sub> 124-7°.

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*Bakanova, Z. M., Mandel'baum, Ya. A. ...*

1.0591, 1.4840; *Et, Et, Et*, 52.5, *b<sub>7-18</sub>* 108-13°, 1.0301,  
 1.4807; *Et, Et, iso-Pr*, 65, *b<sub>7-18</sub>* 112-20°, 1.1118, 1.4867;  
*iso-Pr, Et, iso-Pr*, 39.2, *b<sub>6-12</sub>* 113-20°, 1.0560, 1.4820; *Bu,*  
*Et, iso-Pr*, 70.5, *b<sub>6-12</sub>* 126-40°, 1.0718, 1.4890; *Bu, H, Et,*  
 80.6, *b<sub>6-12</sub>* 100°, 1.2523, 1.5000. (RO)<sub>2</sub>P(S)SCHMeNR'  
 CO<sub>2</sub>R": *Et, H, Et*, 44, *b<sub>6-12</sub>* 74-83°, 1.1592, 1.4896; *iso-Pr,*  
*H, Et*, 39.6, *b<sub>6-12</sub>* 90-3°, 1.1008, 1.4744; *iso-Bu, H, Et*, 43.3,  
*b<sub>6-12</sub>* 95-114°, 1.0845, 1.4906; *Me, Et, Et*, 25.4, *b<sub>7-18</sub>* 70-1°,  
 1.1826, 1.4980; *Pr, Et, Et*, 18.8, *b<sub>6-12</sub>* 75-85°, 1.0703,  
 1.4925; *iso-Pr, Et, Et*, 33, *b<sub>7-18</sub>* 99°, 1.0703, 1.4780; *iso-Bu,*  
*Et, Et*, 61, *b<sub>6-12</sub>* 92-103°, 1.0554, 1.4855; *Me, Et, iso-Pr*, 30,  
*b<sub>6-12</sub>* 65-75°, 1.0595, 1.4973. The substances are said to be  
 weak contact insecticides, but unspecified ones have fairly  
 strong systemic activity.

G. M. Kosolapoff

*7/2*

MEL'NIKOV, N. N.

Organic insectofungicides. XIX. Synthesis of [substituted] ambia-  
 alkyl esters of dithiophosphoric acid. K. D. Shvetsova-Shilovskaya,  
 N. N. Mel'nikov and N. I. Marten'yanova (Zh. obshch. Khim., 1956,  
 28, 193-198).—A series of esters of general formula  
 (OR)<sub>2</sub>PS-S-CH(R'')NR'CO<sub>2</sub>N'' were prepared by the Mannich  
 reaction. R was Me, Et, 1<sup>st</sup>, 1<sup>st</sup>, Bu<sup>o</sup> or Bu<sup>t</sup>; R' was H, Me or  
 Et; R'' was Et or Pr<sup>i</sup>; N'' was H or Me. Physical constants are  
 given for the esters. Most have weak contact insecticidal action,  
 and some, not specified, act as systemic insecticides, approaching  
 pyrophosphoric octamethyltetramide in activity and persistence.  
 R. Tauscov.

3

MELEN'NIKOV, N. N.

✓ Insectofungicides. XX. Synthesis of mixed esters of phosphorothionic acid containing heterocyclic radicals. K. D. Shvetsova-Shlovskaia, N. N. Mel'nikov and A. F. Grapov (*Zh. obshch. Khim.*, 1956, 29, 808-810).—Synthesis is described of esters of phosphorothionic acid with pyrazolone, pyrimidine and furan radicals. Pyrimidine and pyrazolyl esters of phosphorothionic acid were obtained by reaction of diethyl phosphorochloridothionate with K derivatives of heterocyclic compounds by prolonged heating in toluol. Furfuryl diethylphosphothionate was synthesized by treating furfuryl alcohol with diethyl phosphorochloridothionate in pyridine. The compounds obtained exhibited strong insecticide activity. A. L. B.

Chem 3

MEI'NIKOV, N. N.

*(chem)* Organic insectofungicides. XX. Synthesis of mixed esters of thionphosphoric acid containing heterocyclic radicals. K. D. Shvetsova-Shilovskaya, N. N. Mei'nikov, and A. F. Grapov. *J. Gen. Chem. U.S.S.R.* 26, 925-7 (1956)(English translation).—See *C.A.*: 50, 14769g.

3

GM

MEL'NIKOV, N. N.

Organic pesticides. XXI. Synthesis of mixed esters of phosphorodithioic acid. I. L. Vladimirova and N. N. Mel'nikov. XXII. Reactions of dialkyl phosphorochloridates with p-nitrophenol in presence of pyridine hydrochloride. Z. M. Bakanova, Ya. A. Mandel'baum and N. N. Mel'nikov. XXIII. Preparation of dialkylchlorothiophosphites. N. N. Mel'nikov, Ya. A. Mandel'baum, V. I. Lomakin and Z. M. Bakanova. XXIV. New method of preparing halogenophenyl esters of sulphonic acids. I. G. Wallson, S. D. Volodkovich, N. N. Mel'nikov and I. M. Rublava. XXV. Synthesis of mixed esters of phosphorothioic acid. Ya. A. Mandel'baum, N. N. Mel'nikov and V. I. Lomakina (*Zh. obshch. Khim.*, 1956, 20, 2569-2573, 2575-2577, 2577-2579, 2579-2591, 2591-2593).—XXI. Forty new compounds, of which some are claimed to be strong insecticides, are synthesized by combining different unsaturated compounds by means of double bonds with dialkyl (I) and diaryl phosphorodithioates (II), yielding mixed esters of phosphorodithioic acid. The combination of I with esters of chloromaleic, citraconic and itaconic acid went contrary to the Markownikoff rule. In some cases reactions took several weeks. A higher degree of purity was achieved by repeated washing of reaction products with 10% aq.  $\text{Na}_2\text{CO}_3$  and then drying by means of anhyd.  $\text{CaCl}_2$ . Filtered deposits were dissolved in benzene and chromatographed on activated  $\text{Al}_2\text{O}_3$ . The chloromaleic anhydride used was obtained by chlorination of maleic anhydride at high temp. using iron chloride.

XXII. The reaction mechanism of dialkyl phosphorochloridates with p-nitrophenol in the presence of pyridine hydrochloride is discussed; the main product was S-alkyl O,O'-4,4'-dinitrodiphenyl phosphorothioate, which was obtained through disproportionation and re-esterification.

MEL'NIKOV, N. N.

9

Organic insectofungicides. XXI. Synthesis of mixed esters of dithiophosphoric acid. J. L. Vladimirova and N. N. Mel'nikov. *J. Gen. Chem. U.S.S.R.* 26, 2861-6 (1960) (English translation). XXII. Reaction of dialkyl chlorothiophosphates with *p*-nitrophenols in the presence of pyridine hydrochloride. Z. M. Bakanova, Ya. A. Mandel'baum, and N. N. Mel'nikov. *Ibid.* 2867-8. XXIII. Preparation of dialkyl chlorothiophosphates. N. N. Mel'nikov, Ya. A. Mandel'baum, V. I. Lomakina, and Z. M. Bakanova. *Ibid.* 2871-3. XXIV. New method of preparation of halophenyl esters of sulfonic acids. L. G. Vol'fon, S. D. Volockovich, N. N. Mel'nikov, and I. M. Rubleva. *Ibid.* 2875-6. XXV. Synthesis of mixed esters of thiophosphoric acid. Ya. A. Mandel'baum, N. N. Mel'nikov, and V. I. Lomakina. *Ibid.* 2977-8. — See *C.A.* 51, 1824a. B.M.F.

11

GMB

USSR/Organic Chemistry - Synthetic Organic Chemistry, E-2

Abst Journal: Referat Zhur - Khimiya, No 1, 1957, 398

Author: Nametkin, S. S., Mel'nikov, N. N., and Bokarev, K. S.

Institution: None

Title: None

Original

Periodical: Zh. prikl. khimii, 1956, Vol 29, No 3, 459-463

Abstract: The synthesis of  $\beta$ -indolylacetic acid (I) by the reaction of indole (II) with ethyl diazoacetate (III) with subsequent hydrolysis of the ethyl ether of  $\beta$ -indolylacetic acid (IV) (Jackson and Manske, Can. J. Res., Sect. B, 1935, 13, 170) has been studied. It was established that the most suitable catalysts for the reaction of I with II are Cu and  $Cu_2Cl_2$ ; benzene was found to be the best medium. To a boiling solution of 0.97 moles of II in 250 ml  $C_6H_6$  with 0.1-0.25 gm  $Cu_2Cl_2$  add dropwise a solution of 0.97 moles of III in 300 ml  $C_6H_6$ ; reflux 1-1.5 hours; filter after cooling. The solvent is distilled off and IV is obtained in yields of 74.5%, bp 213-225<sup>o</sup>/12 mm. Hydrolysis of IV with 16-17% NaOH yields I; the yield is 82-90%.

Card 1/1

*Mel'nikov, N. N.*

*100*

*Chem* Preparation of heparin. S. S. Naimenin, N. N. Mel'nikov, and K. S. Bokarev. *J. Appl. Chem, U.S.S.R.* 39, 497-501 (1956) (Engl. translation).—See *C.A.* 50, 13867c. *B.M.R.* *3*

*PM*

VOL'PKOVICH, S.I., akademik; MEL'NIKOV, N.N., professor.

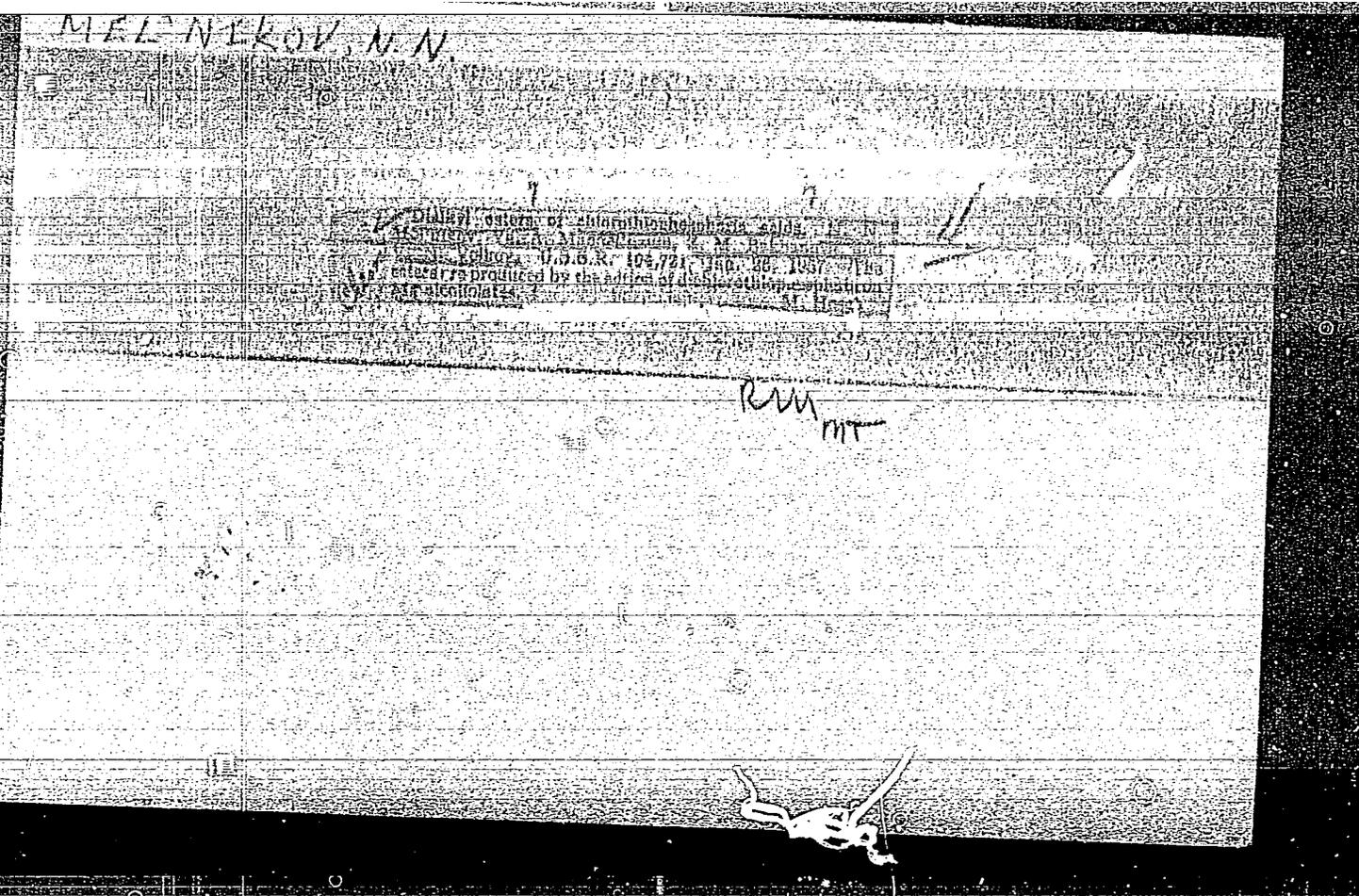
Chemistry in the struggle for high crop yields. Priroda 45 no.2:  
23-37 F '56. (MLRA 9:5)  
(Agricultural chemistry)

1957-1960

MELENIKOV, N. N.

Insecticide. N. N. Mel'nikov, K. D. El'vetsova-Shilovskaya, E. A. Pskrovskii, and A. S. Sedykh. U.S.S.R. 104,641, Jan. 25, 1967. Mixed esters of dithiophosphoric acid contg. a carbamide group in the aliphatic radical, and particularly compds. of the type  $((RO)_2PSSCH_2)_nNCO_2Et$  and  $(RO)_2PSSCH_2N(Me)CO_2Et$  are used as insecticides. M. Hoehl

4



MEF. NIKOV, N. N.

Esters of dithiophosphoric acid. N. N. Melnikov and  
A. E. Grancy. U.S.S.R. 168,423, Apr. 25, 1957. Esters  
of the general type (R<sub>2</sub>)P(S)<sub>2</sub>Ar are obtained by treating  
alkali salts of acid esters of dithiophosphoric acid with  
disorganonitrils. The reaction is carried out in the presence  
of Cu<sub>2</sub> or CuSO<sub>4</sub> as catalyst. M. Hoch

3  
4E47

fm  
10/7

БЕЛНТКОВА, И. И.

3  
pesticides. N. N. Melnikov, Ya. S. Maslov, and  
M. Bakanov, and V. Popov. U.S.P. 2,853,353,  
May 25, 1957. To protect plants from insects and mites,  
dusts or emulsions made with O-methyl O-ethyl O-4-nitro-  
phenyl thiophosphate are used. M. Hosh

MEL'NIKOV, N.N.

Distr: 4E4j

✓ Bisethylxanthogen trisulfide. N. N. Mel'nikov and  
M. L. Galasitina. U.S.S.R. 108,263, Oct. 23, 1957. The  
title compd. is obtained by the action of  $S_2Cl_2$  or  $SCl_2$  on  
alkali metal ethylxanthogenate. M. Hosen //

MEL'NIKOV, N. N. (NIUIF im.Ya. V. Samoylov, Moscow)

"Research works of NIUIF in the field of Organophosphorus Insecticides" (Polskovyye raboty NIUIF v oblasti fosfororganicheskikh insektitsidov)

Chemistry and Uses of Organophosphorus Compounds  
(Khimiya i primeneniye fosfororganicheskikh soyedneniy),  
Trudy of First Conference, 8-10 December 1955, Kazan,  
PP. Published by Kazan Affil. AS USSR, 1957

50-61

Synthesis of organophosphorus compounds described in article carried out by Ya. A. Mandel'baum, K. D. Shvetsova-Shilovskaya, I.L. Vladimirova, A. F. Grapov, N. I. Martem'yanova, Z. M. Bakanova, and V. I. Lomakina; and the testing by P.V. Popov, K. A. Gar, Ye. A. Pokrovskiy, and co-workers.

MEL'NIKOV, N. N.; MANDEL'BAUM, Ya. A. (NIIEF im. Ya. V. Samoylov, Moscow)

"On the Production of Dialkylchlorothiophosphates" (O poluchenii dialkilkhlorotiofosfatov)

Chemistry and Uses of Organophosphorous Compounds  
(Khimiya i primeneniye fosfororganicheskikh sovedneniy),  
Trudy of First Conference, 8-10 December 1955, Kazan,  
pp. Published by Kazan Affil. AS USSR, 1957  
185-193

TERENT'YEV, A.P.; YANOVSKAYA, L.A.; RUKHADZE, Ye.G., redaktor;  
RODIONOV, V.M., akademik, redaktor [deceased]; KAZANSKIY, B.A.,  
akademik, redaktor; KNUNYANTS, I.L., akademik, redaktor;  
SHEMYAKIN, M.M., redaktor; MEL'NIKOV, N.N., prof, redaktor;  
LUR'YE, M.S., tekhnicheskij redaktor.

[Polarographic analysis in organic chemistry] Poliarograficheski  
method v organicheskoi khimii. Moskva, Gos. nauchno - tekhn. izd.  
vo khim. lit-ry, 1957. 388 p. (Reaktsii i metody issledovaniia  
organicheskikh soedinenii, vol.5) (MIRA 10:10)

1. Chlen-korrespondent AN SSSR (for Shemyakin).  
(Polarography) (Chemistry, Organic)

MEL'NIKOV, N. N.

4E300

A new method for the preparation of mixed esters of di-  
 phosphoric acid. N. N. Mel'nikov, A. P. Rumayev, and  
 G. I. Shvet'saya-Shtolovskaya. *Khimi. Nauka i Prom.* 2:  
 204-6 (1957). — Aq. (RO)<sub>2</sub>PSSAr was added dropwise to Ar'  
 N<sub>2</sub>X (obtained by diazotization of the appropriate amine  
 with NaNO<sub>2</sub> at 0°) in the presence of Cu. The mixt. was  
 heated to 60-80° to stoppage of N<sub>2</sub> evolution, extd. with dil.  
 NaOH, dried, and distd. in vacuo. Esters with the following  
 R and Ar were thus prepared: Me, Ph and C<sub>6</sub>H<sub>5</sub>CH<sub>2</sub>-Et, Et,  
 C<sub>4</sub>H<sub>9</sub>Me, C<sub>2</sub>H<sub>5</sub>Cl, C<sub>6</sub>H<sub>5</sub>NO<sub>2</sub>, C<sub>6</sub>H<sub>5</sub>CO<sub>2</sub>Me, and C<sub>6</sub>H<sub>5</sub>OMe.  
 I. Bencosid

MEL'NIKOV, N.N.

Chemistry in the service of plant protection. Zashch. rast. ot  
vred. i bol. 2 no.6:41-44 N-D '57. (MIRA 16:1)  
(Agricultural chemicals)

MEL'NIKOV, N. N.

64 - 7 - 6/12

AUTHORS: Mel'nikov, N. N., Rokhlin, M. I.

TITLE: Chemicals for the Preservation of Plants  
(Khimicheskiye sredstva zashchity rasteniy).

PERIODICAL: Khimicheskaya Promyshlennost', 1957, Nr 7,  
pp. (417)35 - (421)37, (USSR)

ABSTRACT: Examples are first given how to deal with pests and diseases by chemical means. The basic methods for the practical application of chemical preparations are then described. A survey is given of research institutes: NIUIF - Scientific Research Institute for manures as well as for means of destroying insects and mushrooms, the Scientific Research Institute for half-finished goods and dyes, the State Institute for the nitrogen industry, the institute for organic chemistry, the institute for element-organic compounds, the Kazhan branch AN USSR, the academy of agricultural sciences, etc. There follows a detailed description of the production of DDT, which, at present, contains up to 76 % dichloridophanyltrichloromethylmethane, as well as of the various varieties of this preparation. A description and a survey is given of

CARD 1/3

64 - 7 - 6/12

Chemicals for the Preservation of Plants

the production of the second-important insecticide, hexachlorine-cyclohexane, by means of which it was possible to destroy locusts in the USSR completely. Although the method of photochemical chlorization of benzene, which is employed also in the U.S.A. and in Western Germany, must be considered to be the best at present, a far more rational method for the production of a preparation containing 13-35 % gamma isomer was developed in the USSR. The preparation is also transformed into "lindan" (100% gamma isomer), so that it loses its unpleasant smell. Since 1942 chlorinated turpentine, the emulsion concentrate of which is used as SK9 preparation even today, is being produced. In the U.S.A. a similar preparation, which contains a little more chlorine, and is known as "Stroban" has been in use only since a very short time. There follows a survey of the various insecticides etc. with contact- and system effect of the various methods and preparation for the pickling

CARD 2/3

*Mel'nikov N.N.*

64-8-16/19

AUTHOR: Mel'nikov, N. N.

TITLE: Fourth International Congress for Plant Protection  
(Chetvertyy mezhdunarodnyy kongress po zashchite  
rasteniy).

PERIODICAL: Khimicheskaya Promyshlennost', 1957, Nr 8, pp. 51-52 (USSR)

ABSTRACT: The congress took place in Hamburg from September 8<sup>th</sup> up to  
September 15<sup>th</sup>, 1957. 60 countries, among them also the  
USSR, took part in it. Approximately 1200 persons were  
present. More than 420 speeches were delivered. A short  
general review about the themes dealt with are given without  
the titles of the reports and without the names of the  
reporters.

AVAILABLE: Library of Congress

Card 1/1

MEL'NIKOV, N.N.

Fourth International Congress on Plant Protection. Khim. prom. no.8:  
499-500 D '57. (MIRA 11:2)  
(Hamburg--Plants, Protection of--Congresses)

MELNIKOV, N. N.

8  
4E4 i  
4E3 i  
4E2 C (j)  
2 May

Organic insectofungicides. XXVI. New method of synthesis of mixed esters of dithiophosphoric acid. N. N. Mel'nikov, A. F. Granov, and K. D. Shvel'sova-Shilovskaya. (Sci. Inst. Fertilizers and Insectofungicides, Moscow). *Zhur. Obshch. Khim.* 27, 1905-7 (1957); *Ch. C.A.* 51, 1824a. The following method was used to prep. mixed esters of dithiophosphoric acid. To an aq. soln. of an aryl diazonium salt, freed of excess HNO<sub>3</sub> by air-blowing, was added an aq. soln. of equimolar amt. of (RO)<sub>2</sub>PS<sub>2</sub>H in the form of a salt, such as K, and 0.1 g. powd. Cu/0.1 mole reactants. The mixing was done at 0-2° and the mixt. warmed to 40-60° until N evolution terminated. After cooling to 10-15°, the mixt. was extd. with Et<sub>2</sub>O and the dried ext. distd. Thus were obtained: PhSP(S)(OMe)<sub>2</sub>, 51%, b.p. 95-7°, d<sub>4</sub> 1.2460, n<sub>D</sub> 1.5927 (gives 50% kill of *Calandra oryzae* at 0.021% in 50 hrs.); PhSP(S)(OEt)<sub>2</sub>, 48%, b.p. 102.5-1.5°, 1.1823, 1.5629 (0.03%); *o*-MeC<sub>6</sub>H<sub>4</sub>SP(S)(OMe)<sub>2</sub>, 40.5%, b.p. 101-2.5°, 1.1614, 1.5660 (-%); *m*-isomer, 27.5%, b.p. 102.5-2.8°, 1.1729, 1.5662 (-%); *p*-isomer, 42%, b.p. 104.5-5°, 1.2130, 1.5829 (0.03%); *o*-MeC<sub>6</sub>H<sub>4</sub>SP(S)(OEt)<sub>2</sub>, 47.5%, b.p. 118-18.5°, 1.1690, 1.5612 (0.07%); *m*-isomer, 42%, b.p. 105-2°, 1.1781, 1.5624 (0.012%); *p*-isomer, 45%, b.p. 104.6-1.5639 (0.035%); *m*-ClC<sub>6</sub>H<sub>4</sub>SP(S)(OEt)<sub>2</sub>, 41.6%, b.p. 6.5°, 1.2538, 1.5730 (-%); *p*-isomer, 51.3%, b.p. 122-3° 1.2627, 1.5750 (-%); *p*-ClC<sub>6</sub>H<sub>4</sub>SP(S)(OEt)<sub>2</sub>, 41.6%, b.p. 137-9°, m. 45.5-6.5° (0.001%). If the soln. of the diazonium salt is not freed of N oxides the (RO)<sub>2</sub>PS<sub>2</sub>H is oxidized to the disulfide and the products are badly contaminated.

G. M. Kosolapoff

*MEL'NIKOV, N. N.*

MEL'NIKOV, N.N.; MANDEL'BAUM, Ya.A.; SVENTSITSKIY, Ye.I.; BAKANOVA, Z.M.

On organic insectofungicides. Part 27: New method for the  
preparation of esters of chlorothiophosphoric acid. Zhur.ob.khim.  
27 no.7:1908-1910 J1 '57. (MIRA 10:10)

1. Nauchnyy institut po udobreniyam i insektofungisidam.  
(Insecticides) (Chlorothiophosphoric acids)

*Mel'nikov, N.N.*  
MELNIKOV, N. N.

"Certain Trends in the Development of Chemistry and the Chemical Industry  
in the U.S.S.R.,"

paper submitted at Chemical Engineering Conf. Montreal, 20-23 Apr 58

TRANS Available; B 3,104,359, 12 May 58

M-5

USSR/Cultivated Plants - Commercial. Oil-Bearing. Sugar-  
Bearing.

Abs Jour : Ref Zhur - Biol., No 20, 1958, 91748

Author : Mel'nikov, N.N., Galashina, M.L.

Inst : -

Title : Studies of New Chemical Methods for the Pre-Harvest  
Removal of Cotton Plant Leaves.

Orig Pub : B sb.: Materialy Ob'yedin. nauchn. sessii po khlopkovo-  
dstvu. T. 2 Tashkent. Gosizdat UzSSR, 1958, 250-256.

Abstract : Nonabstract.

MEL'NIKOV, N.N.

Hydrocarbons, halogen derivatives of hydrocarbons, and nitro com-  
pounds. Itogi nauki: Biol.nauki no.2:59-68 '58. (MIRA 14:4)  
(Hydrocarbons) (Nitro compounds)

MEL'NIKOV, N.N.

Alcohols, phenols, and simple esters. Itogi nauki: Biol.nauki  
no.2:69-78 '58. (MIRA 14:4)

(Alcohol)

(Esters)

(Phenols)

AUTHORS: Baskakov, Yu. A., Mel'nikov, N.N. 64-58-3-9/20  
TITLE: On the Production of Maleic Acid Hydrazide (O poluchenii  
gidrazida maleinovoy kisloty)  
PERIODICAL: Khimicheskaya Promyshlennost', 1958, Nr 3, PP  
32-35 (USSR)

ABSTRACT:

The growth-inhibiting property of cyclic anhydride of maleic acid caused an intensive investigation of this substance, mainly in 1949 - 1954. This compound has three tautomeric forms, the crystalline form being the oxy-pyridazone form. With lye the hydrazide behaves like a monobasic acid. The Na-, K-, and ammonium salts in this case are well soluble in water. Different reactions of the hydrazide are mentioned as well as some specific properties with regard to its effect on plants and warm-blooded animals which were taken from publications. With that also the first syntheses of Curtius (reference 52) and of Arndt, Loewe, and Ergener (reference 53) are mentioned. In order to investigate the production technology a systematical investigation of the processes in presence of different organic and inorganic acids was made. With that a strong

Card 1/2

On the Production of Maleic Acid Hydrazide

64-58-3-9/20

dependence of the hydrazide yield on the pH of the medium was stated, and high yields are obtained with linear molecular acids the dissociation constant of which is greater than that of maleic acid. The formation of hydrazide takes place in two stages, first an acid hydrazide is formed and then it is cyclized. For the production of salts of maleic acid hydrazide the following scheme is proposed: production from hydrazine sulfate and anhydride of maleic acid in an aqueous medium with warming up and mixing. Cooling of the reaction mixture. Filtration and washing of the hydrazide with subsequent drying and production of salts from maleic acid hydrazide and diethanolamine or sodium lye. There are 2 tables and 58 references, 11 of which are Soviet.

ASSOCIATION: Institut fiziologii rasteniy imeni K. A. Timiryazeva AN SSSR  
(Institute for Plant Physiology imeni K. A. Timiryazev,  
AS USSR)

1. Malic acid hydrazide--Production
2. Malic acid hydrazide--Physiological effects
3. Malic acid hydrazide--Chemical properties
4. Herbicides--Analysis

Card 2/2

BASKAKOV, Yu.A.; MEL'NIKOV, N.N.

Preparation of maleic acid hydrazide. Khim. prom. no. 3:160-163  
Ap-My '58. (MIRA 11:6)

1. Institut fiziologii rasteniy imeni K.A. Timiryazeva AN SSSR.  
(Maleic acid)

MEL'NIKOV, N.N., prof., doktor khim.nauk

Phosphorus organic insecticides and acaricides. Zashch. rast.  
ot vred. i bol. 3 no.5:13-15 S-0 '58. (MIRA 11:10)  
(Insecticides) (Phosphorus organic compounds)

BASKAKOV, Yu.A.; MEL'NIKOV, N.N.

New herbicides produced from phenyl-hydroxylamine. *Khim.nauk i*  
prom. 3 no.5:683-684 '58. (*MIRA* 11:11)

1. Nauchno-issledovatel'skiy institut udobreniy i insektofungitsidov.  
(Hydroxylamine) (Carbamic acid) (Herbicides)

MEL'NIKOV, N. N.

79-1 32/63

AUTHORS:

Kukalenko, S. S., Mel'nikov, N. N.

TITLE:

**Organic Insecti- and Fungicides** (Iz oblasti organicheskikh insektofungitsidov). XXVIII. The Synthesis of Some Ethers of Bicyclo-(2,2,1)-Heptenyl-5-Carbinol-2 (XXVIII). Sintez nekotorykh efirov bitsiklo-(2,2,1)-septenil-5-karbinola-2).

PERIODICAL:

Zhurnal Obshchey Khimii, 1958, Vol. 28, Nr 1, pp. 154-157 (USSR).

ABSTRACT:

Much attention was recently paid to different polycyclic compounds obtained by the way of diene-synthesis from cyclopentadiene and hexachlorocyclopentadiene, because extremely active insecticides were found among them (reference 1-6). For this reason syntheses of similar polycyclic compounds were performed, in order to determine the dependence of this activity on their structure. Above all the authors synthesized various ethers of bicyclo-(2,2,1)-heptenyl-5-carbinol-2 by conversion of cyclopentadiene with allyl esters of different acids. These syntheses were of importance, as the condensation of cyclopentadiene was hitherto known in publications only with allyl alcohol, chloro- and bromoallyl, some vinyl ether, acrolein.

Card 1/3

## Organic Insecti- and Fungicides

XVIII The synthesis

79-1 32/83

of Some Others of Bicyclo-(2,2,1)-heptene-5 Carbons-2.

acetylene and a few other compounds. The reaction of cyclopentadiene with the allyl esters of various acids takes place at an elevated temperature in an autoclave or in sealed tubes at some pressure. The conversion can be represented according to the given scheme. The synthesized compounds and their properties are described in the table. As in the published syntheses of cyclopentadiene with allyl alcohol, bromo- and chloro-allyl only the boiling points were given, the authors also described the syntheses of these compounds, as well as the synthesis of 2,3-dichlorobicyclo-(2,2,1)-heptene-5 from trans-dichloroethylene and cyclopentadiene. For the purpose of a synthesis of the different arylbicyclo-(2,2,1)-heptadienes the conversion of cyclopentadiene with phenylchlorophenyl-, bromophenyl- and tolylacetylenes was investigated, but it was not possible to isolate the monomeric reaction products. Only solid polymers with high molecular weights were obtained on that occasion. There are 1 table and 13 references 10 of which are Slavic.

ASSOCIATION:  
Card 2/3

Scientific Institute for Fertilizers, Insecticides and Fungicides (Nauchnyy institut'po udobreniyam i insektofungitsidam).

**Organic Insecti- and Fungicides**

of Some Ethers of Bicyclo-(2,2,1)-Heptenyl-5-Carbinol-2).

XXVIII. The Synthesis

79-1-32/63

SUBMITTED: December 18, 1956

AVAILABLE: Library of Congress

Card 3/3

1. Chemistry 2. Ethers 3. Cyclic compounds

MEL'NIKOV, N. N.

AUTHORS: Kuznetsov, S. S., Mel'nikov, N. N.

79-1-33/83

TITLE: **Organic Insecti- and Fungicides** (Iz oblasti organicheskikh i insektofungitsidov). XXIX. On the Interaction of Hexachlorocyclopentadiene With Some Unsaturated Compounds (XXIX. O vzaimodeystvii geksakhlortsiklopentadiyena s nekotorymi nepredel'nymi soyedineniyami).

PERIODICAL: Zhurnal Obshchey Khimii, 1958, Vol. 28, Nr 1, pp. 157-161 (USSR).

## ABSTRACT:

In order to find new active insecticides, the authors investigated the reaction products of hexachlorocyclopentadiene with various unsaturated compounds, first of all some ethers of allyl alcohol and bicyclo-(2,2,1)-heptenyl-5-carbinol-2. Moreover the reaction of hexachlorocyclopentadiene with ethyl- and isobutylvinyl-ethers was investigated. The conversion of the allyl esters of different fatty acids and the allyl ester of dithiophosphoric acid with hexachlorocyclopentadiene takes place comparatively easily on heating of equimolecular amounts of the initial product at 100-125° C within 10-12 hours, where the corresponding derivatives are obtained with yields of 50-60%. These conversions can be represented by scheme (I).

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**Organic Insecti- and Fungicides**

XXIX. On the Interaction 73 1 33/63

of Hexachlorocyclopentadiene with Some Unsaturated Compounds.

The synthesized compounds and their properties are given in the table. Some of these compounds are already known from publications, but they were either synthesized according to other methods or their constants were not given. The condensation of hexachlorocyclopentadiene with the ethers of bicyclo-(2,2,1)-heptenyl-5-carbinol-2 takes place considerably slower so that a heating of the reaction mixture at 125-130° C and not less than 15 hours are needed in order to obtain a 50% yield of the derivative. This reaction takes place according to scheme (II). Of all investigated compounds the vinyl ethers react most easily with hexachlorocyclopentadiene and they are bound to the latter already at 80-90° C, but the reaction with the bicyclic compounds even at elevated temperatures takes place considerably slower. As far as the insecticide activity is concerned they have a much smaller effectiveness than this is the case with aldrin and even chlorindane. There are 1 table, and 3 references, 5 of which are Slavic.

ASSOCIATION: Scientific Institute for Fertilizers and Insecticides (Nauchnyy institut po udobreniyam i insektofungitsidam).

Card 2/2

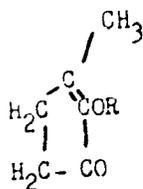
MEL'NIKOV, N. N.

AUTHORS: Mel'nikov, N. N., Shvetsova-Shilovskaya, K. D. 79-2-Lo/64

TITLE: From the Field of Organic Insectofungicides (Iz oblasti organicheskikh insektofungitsidov).  
 XXX. The Synthesis of Some Derivatives of 1-Methylcyclopentandion-2,3  
 (XXX. Sintez nekotorykh proizvodnykh 1-metiltsiklopentandiona-2,3).

PERIODICAL: Zhurnal Obshchey Khimii, 1958, Vol. 28, Nr 2, pp. 474-476 (USSR).

ABSTRACT: The enolesters of the 1-methylcyclopentandion-2,3 as well as of diketone were synthesized and tested for their insectofungicidal-properties. The reactions were started from the 1-methylcyclopentandion and the anhydride of chlorine of the corresponding acids. The compounds of the general formula:



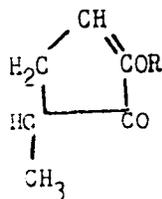
were obtained as reaction products, as well as smaller admixtures of:

Card 1/2

From the Field of Organic Insectofungicides.

XXX. The Synthesis of Some Derivatives of 1-Methylcyclopentandion-2,3.

77-2-10/64



The synthesized compounds and their specific properties as well as the working methods are given. The results of the investigations are the following: only the esters of thiophosphoric acid are effective as insecticide, whereas those of carbamic-, acetic-, butyric-, and carbonic acid were inactive.

There are 1 table, and 2 references, 1 of which is Slavic.

ASSOCIATION. Scientific Institute for Fertilizers, Insecticides and Fungicides (Nauchnyy institut po udobreniyam i insektofungitsidam).

SUBMITTED. January 16, 1957.

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Card 2/2

MEL'NIKOV, N. V.

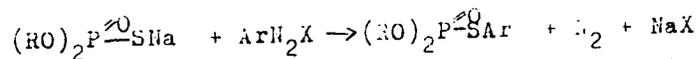
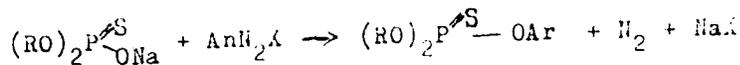
79-3-1/2

AUTHORS: Mel'nikov, N. N., Mandel'baum, Ya. A.,  
Lomakina, V. I.,

TITLE: Organic Insecticides and Fungicides  
 (Iz oblasti organicheskikh insektofungitsidov) XXXI A New method for the pro-  
 duction of mixed esters of Thiophosphoric Acid ( XXXI. Lovj,  
 spsob polucheniya smeshannykh efirov tiofosfornoy kisloty)

PERIODICAL: Zhurnal Obshchey Khimii, 1958, Vol. 28, Nr 2, pp. 476-478  
 (USSR)

ABSTRACT: The reaction between aromatic diazocompounds with salts of the  
 dialkylthiophosphoric acid was investigated and mixed ethers of  
 thiophosphoric acid were obtained in acid-aqueous solution with  
 a yield of up to 50%. Here schematically seen an isomer mixture  
 of the tautomers is formed:



In the present paper the equilibrium of the tautomers of the  
 dialkylthiophosphates is moved to the side of the thioform, in

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